

# **Dynameomics: Protein Mechanics, Folding and Unfolding through Large Scale All-Atom Molecular Dynamics Simulations**

INCITE 6

David A. C. Beck

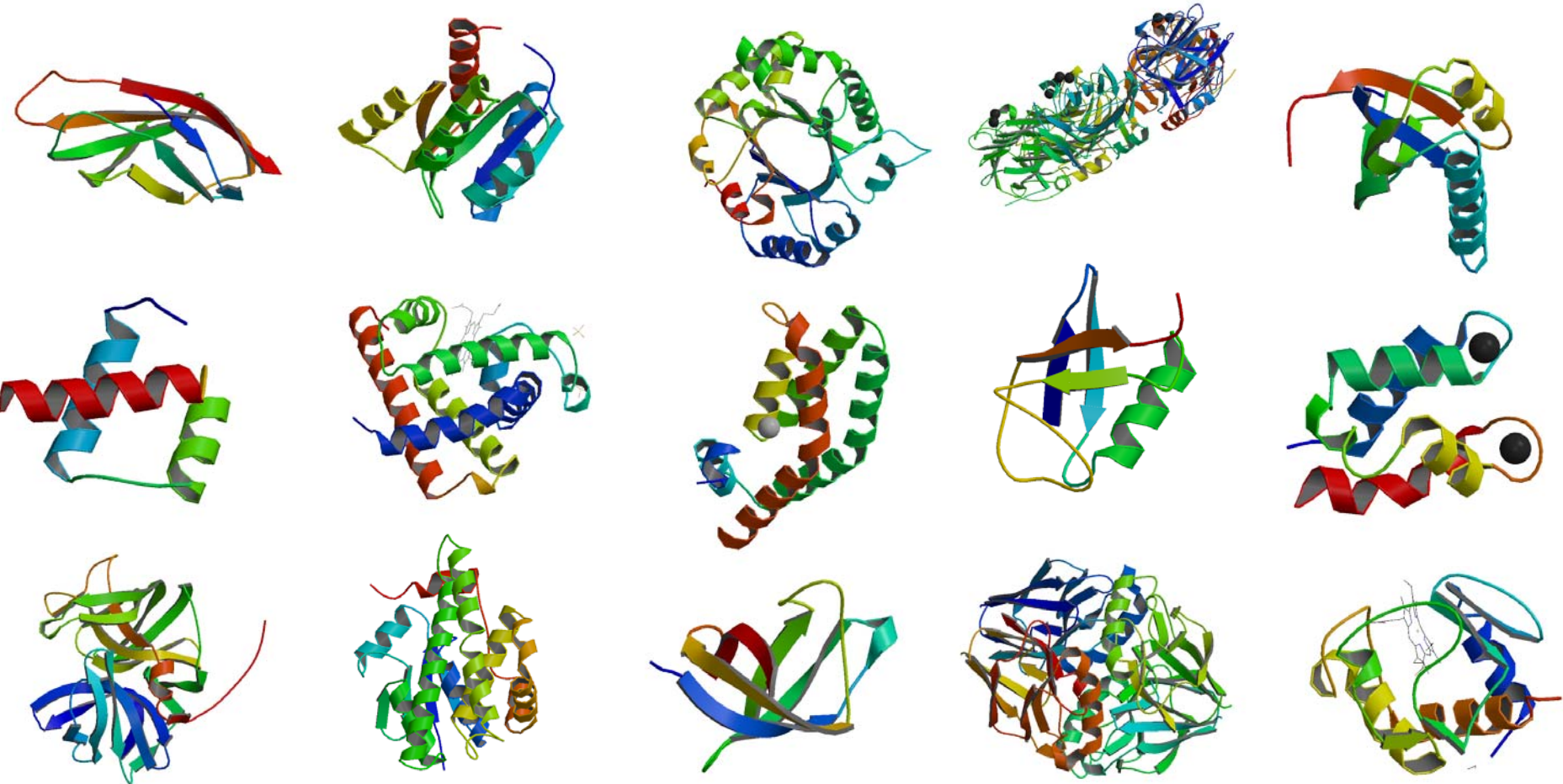
Valerie Daggett Research Group

Department of Medicinal Chemistry  
University of Washington, Seattle  
November 15<sup>th</sup>, 2005

# Proteins

---

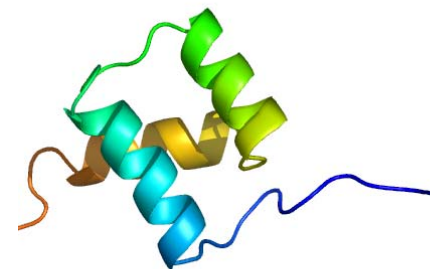
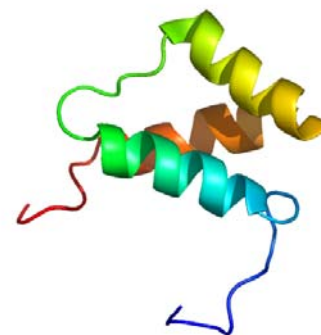
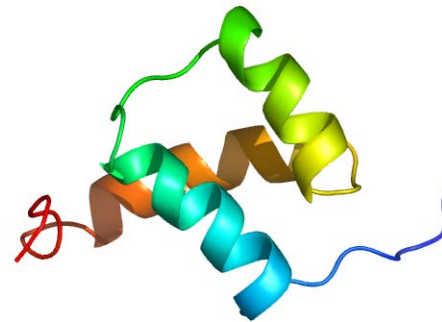
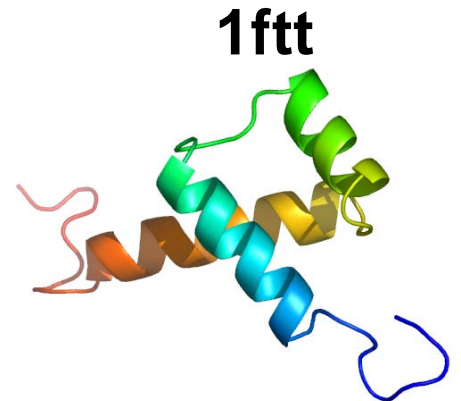
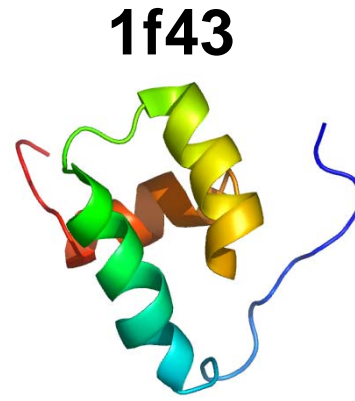
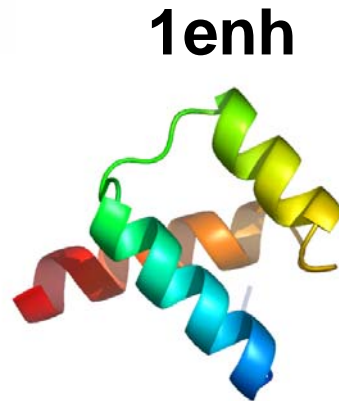
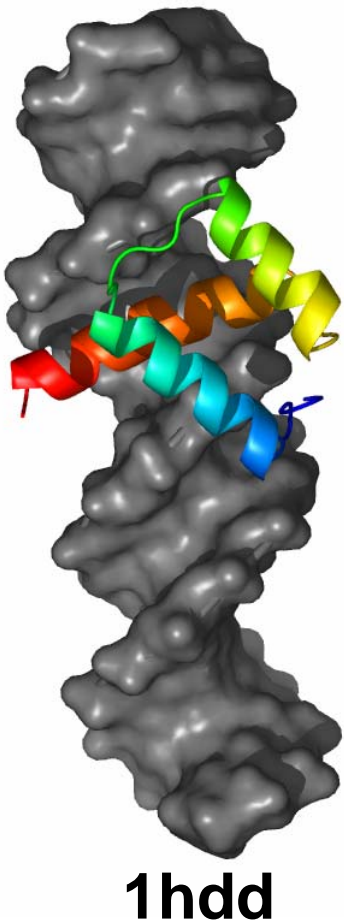
- **Proteins are life's machines, tools and structures**
  - Many jobs, many shapes, many sizes



# Proteins

---

- **Proteins are life's machines, tools and structures**
  - Nature reuses designs for similar jobs



1bw5

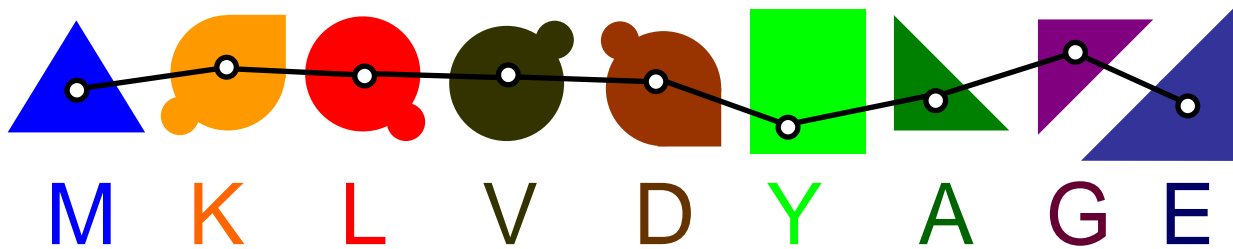
1du6

1cqt

# Proteins

---

- **Proteins are hetero-polymers of specific sequence**

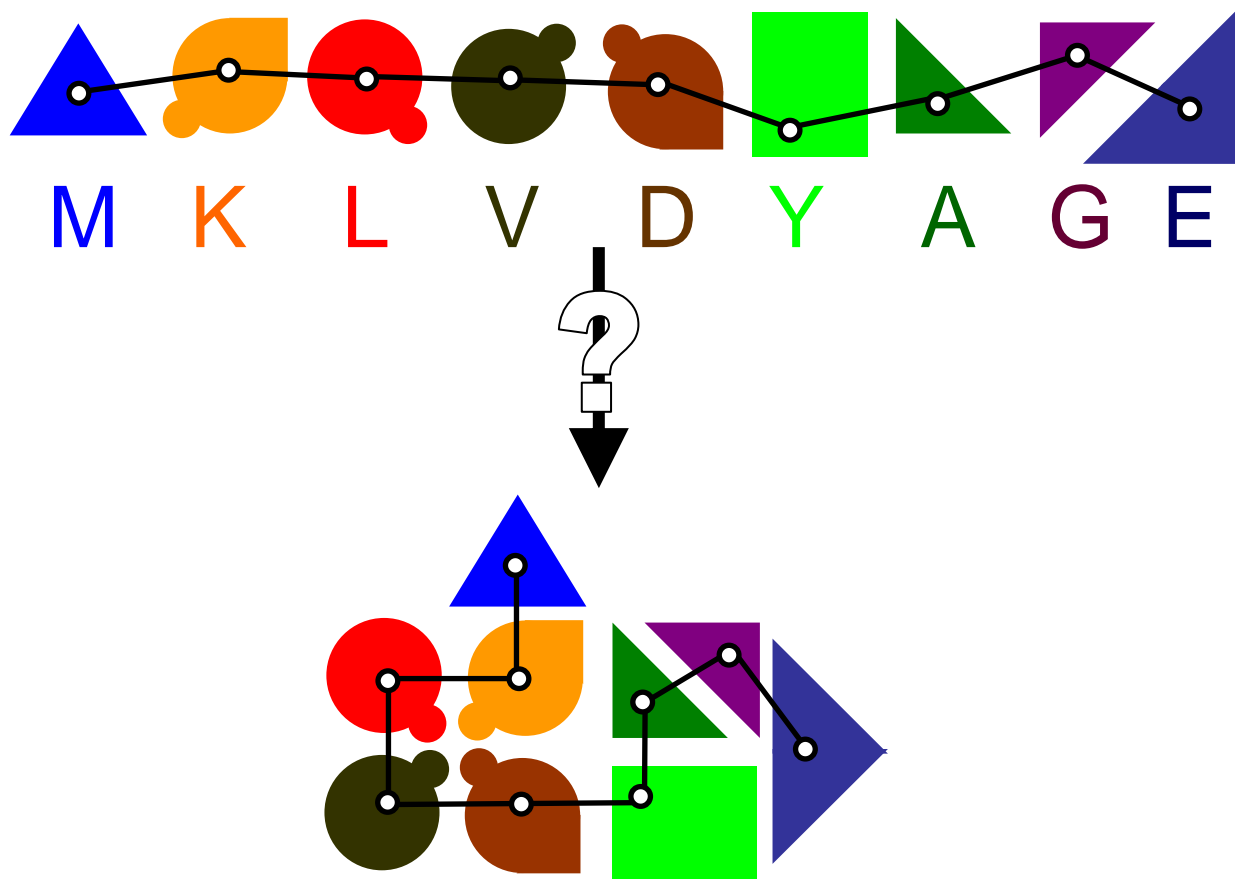


- **There are 20 common polymeric units (amino acids)**
  - **Composed of a variety of basic chemical moieties**
- **Chain lengths range from 40 amino acids on up**

# Proteins

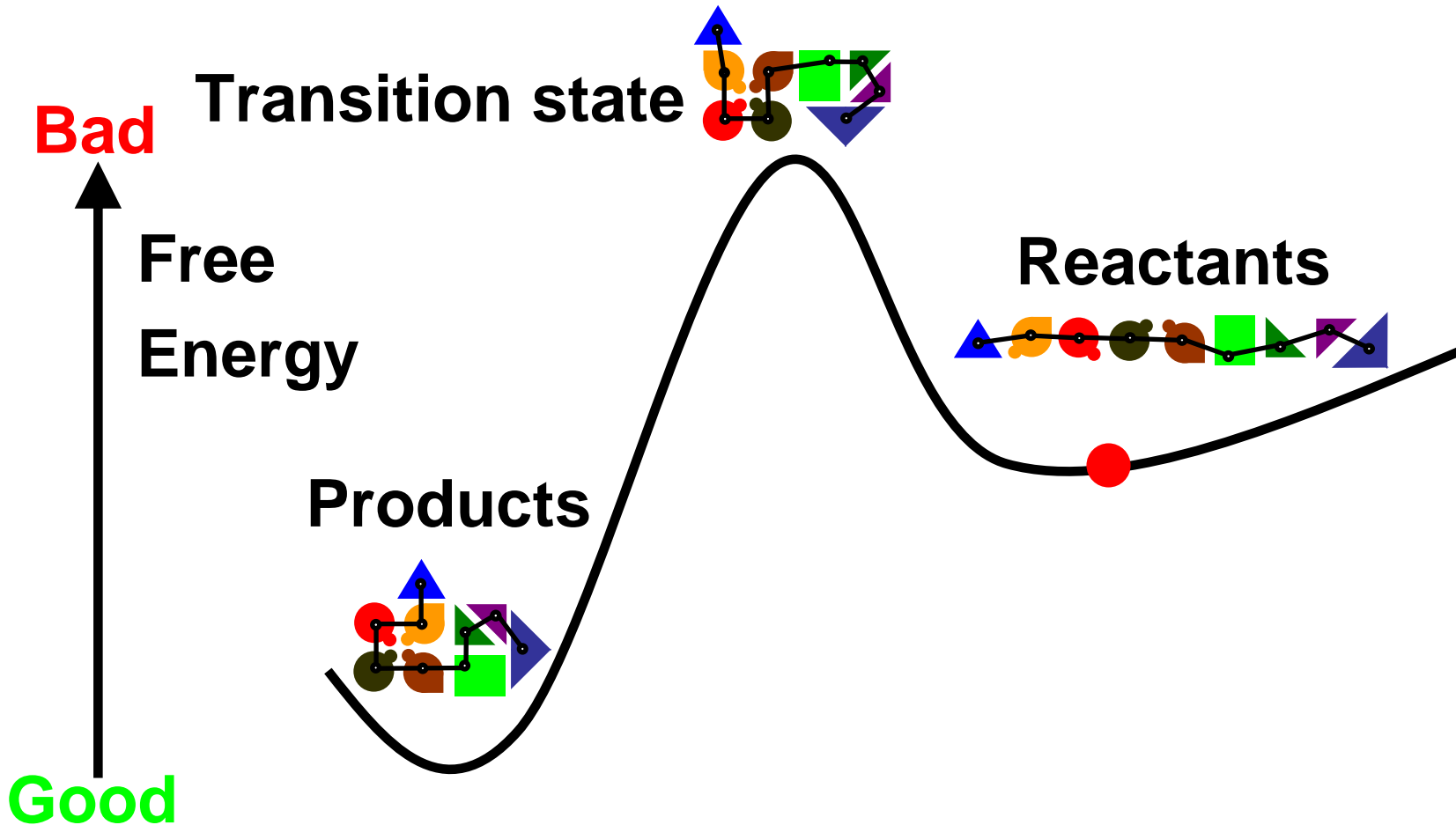
---

- Proteins are hetero-polymers that adopt a unique fold



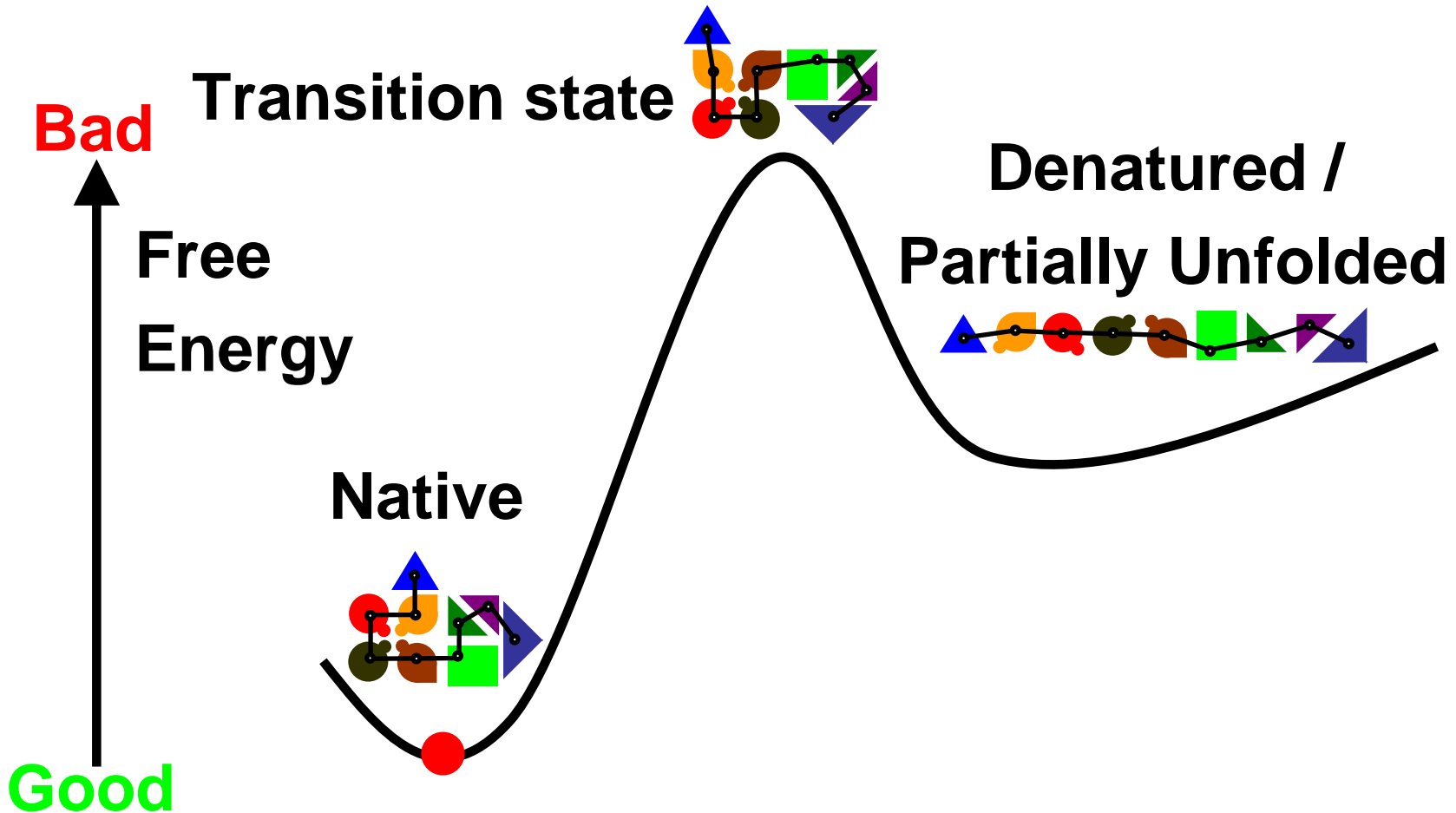
# Proteins

- Protein folding as a reaction



# Proteins

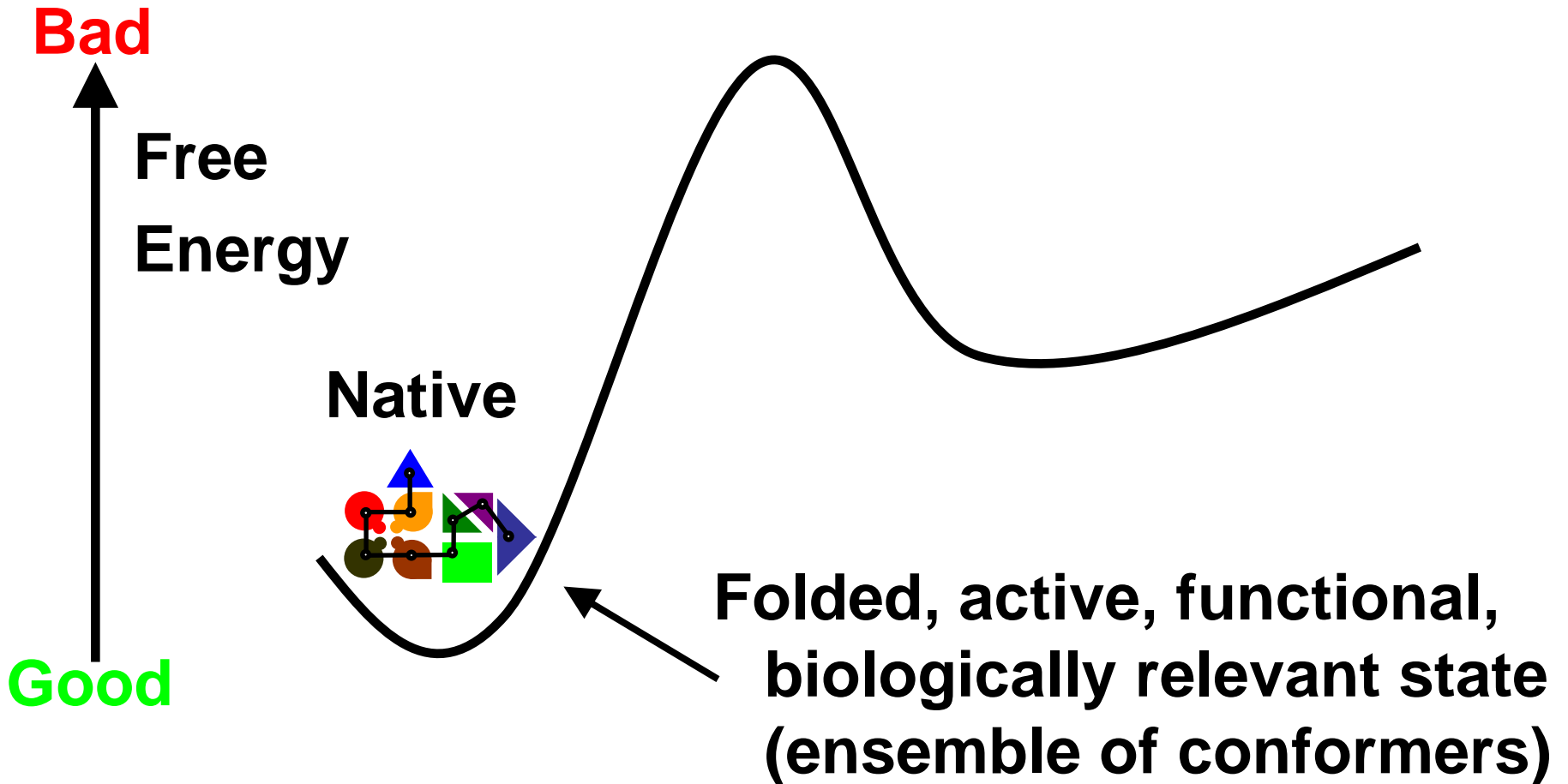
- Protein folding ...



# Proteins

---

- Folded proteins

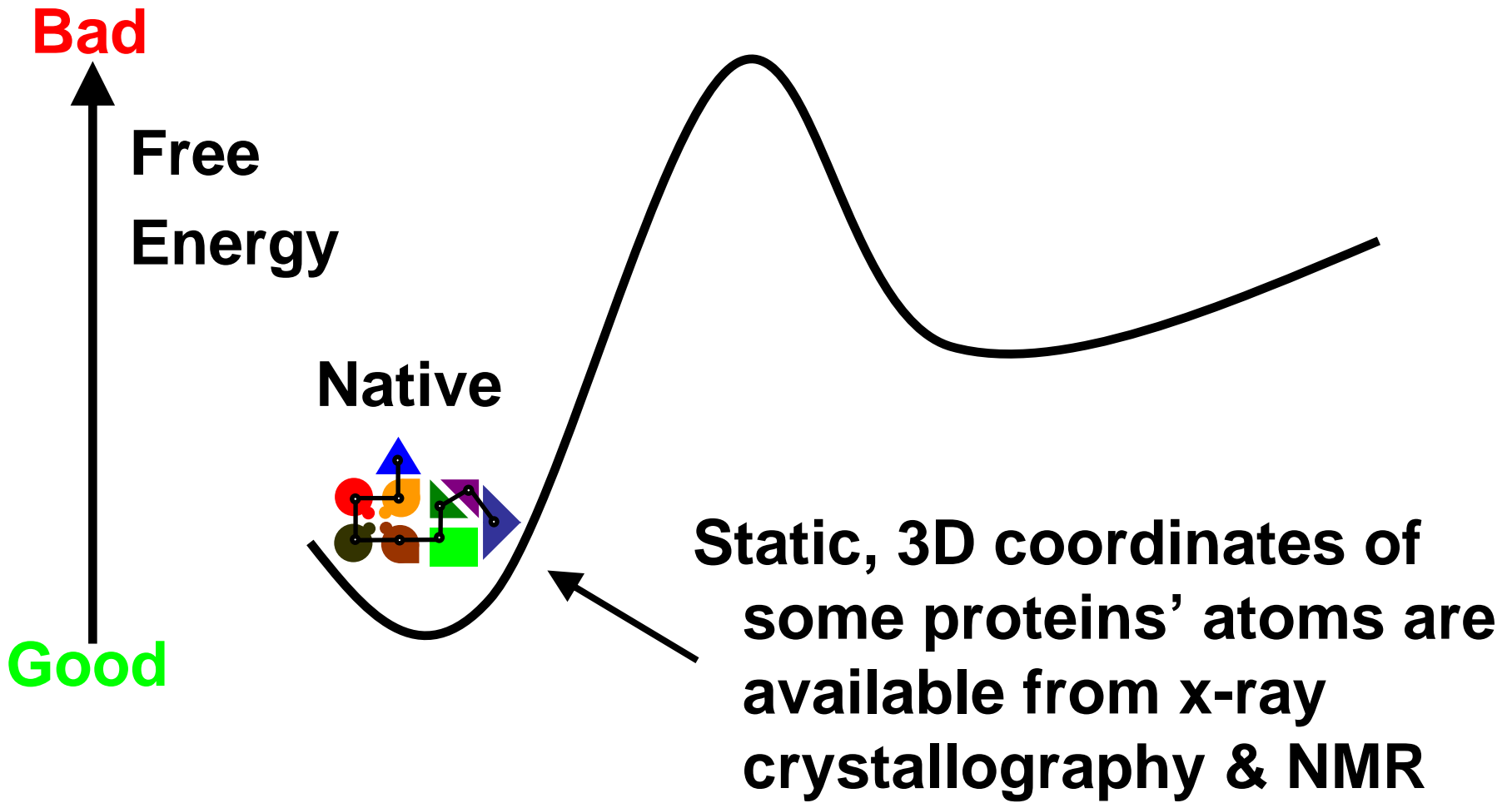




# Proteins

---

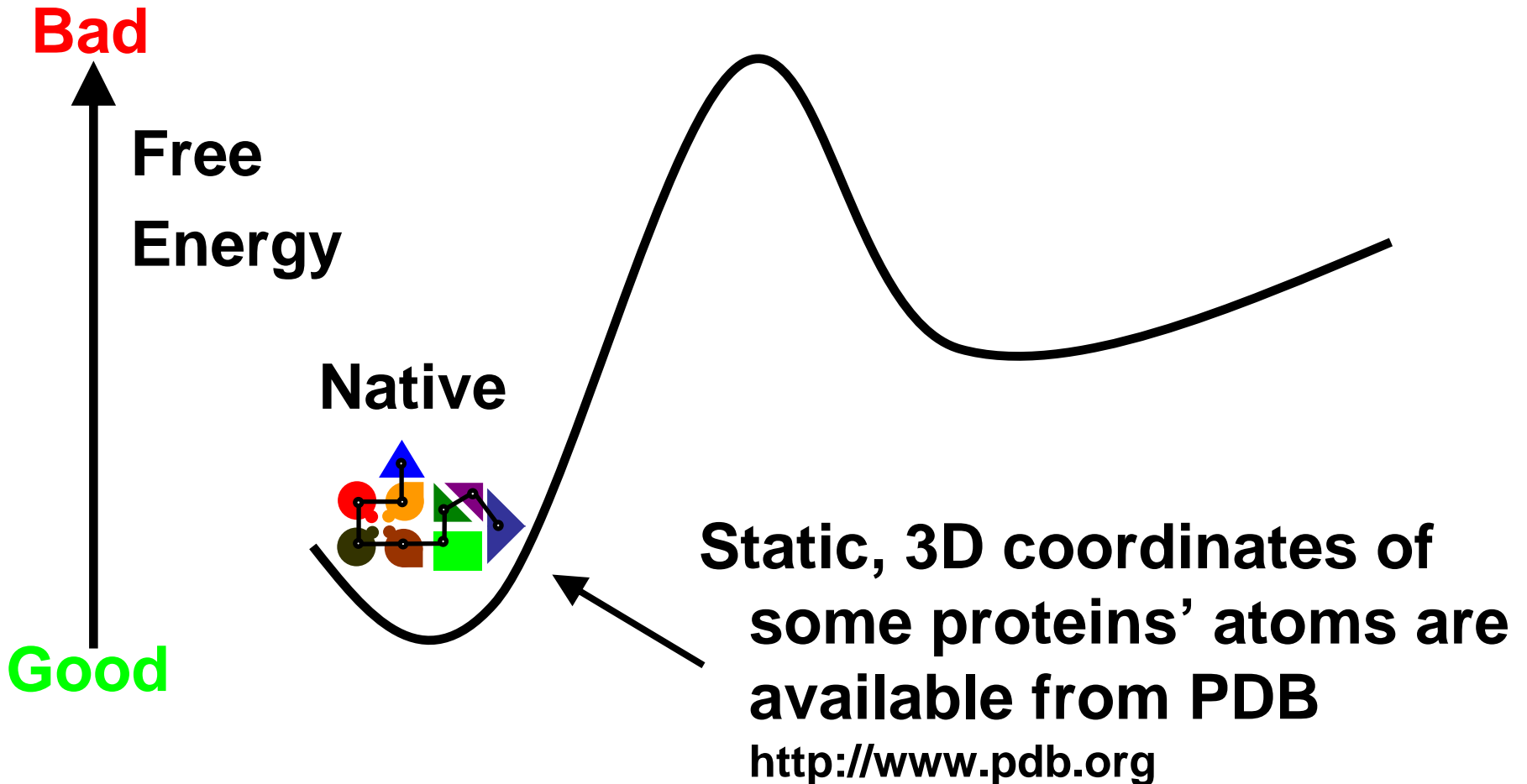
- Folded proteins



# Proteins

---

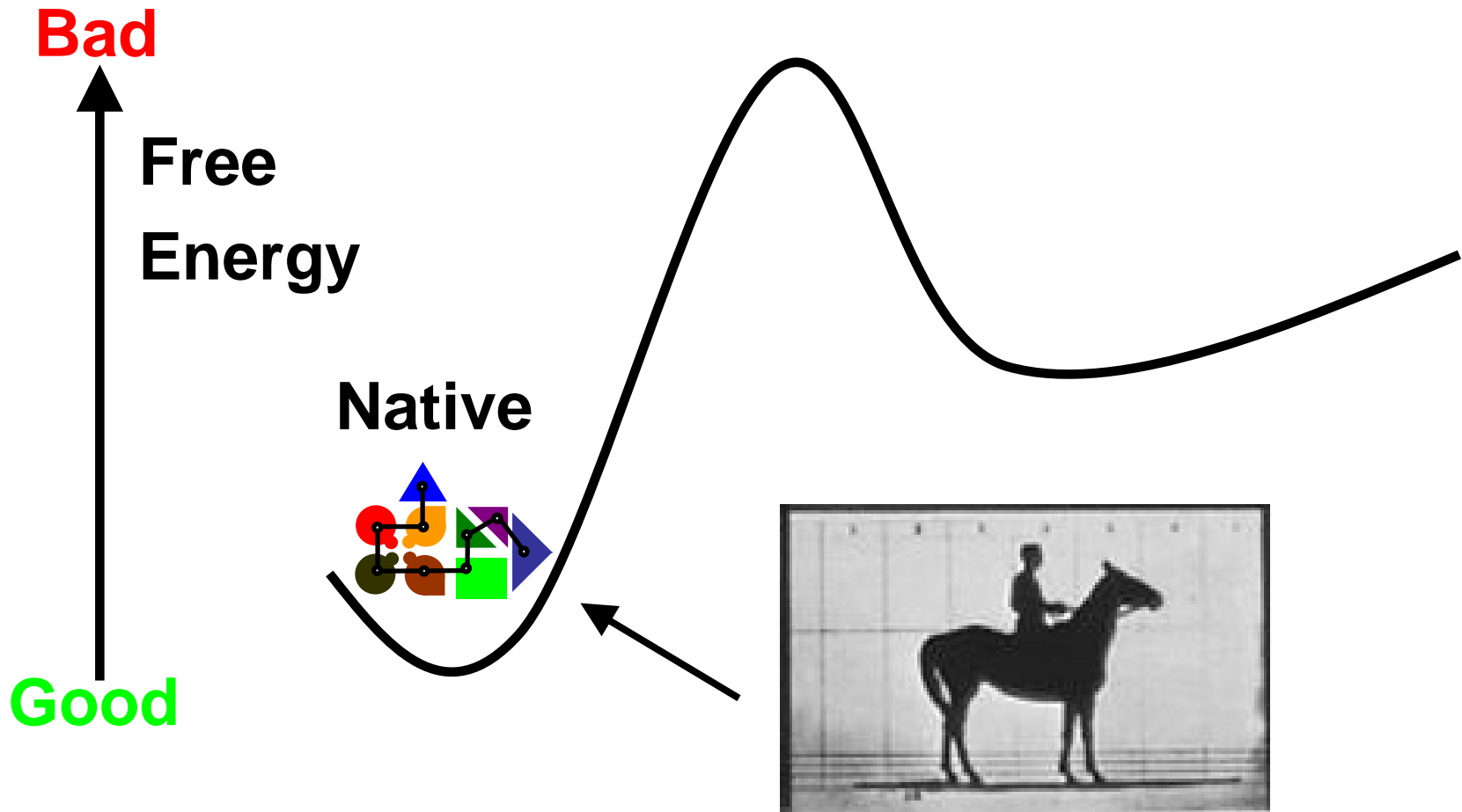
- Folded proteins



# Proteins

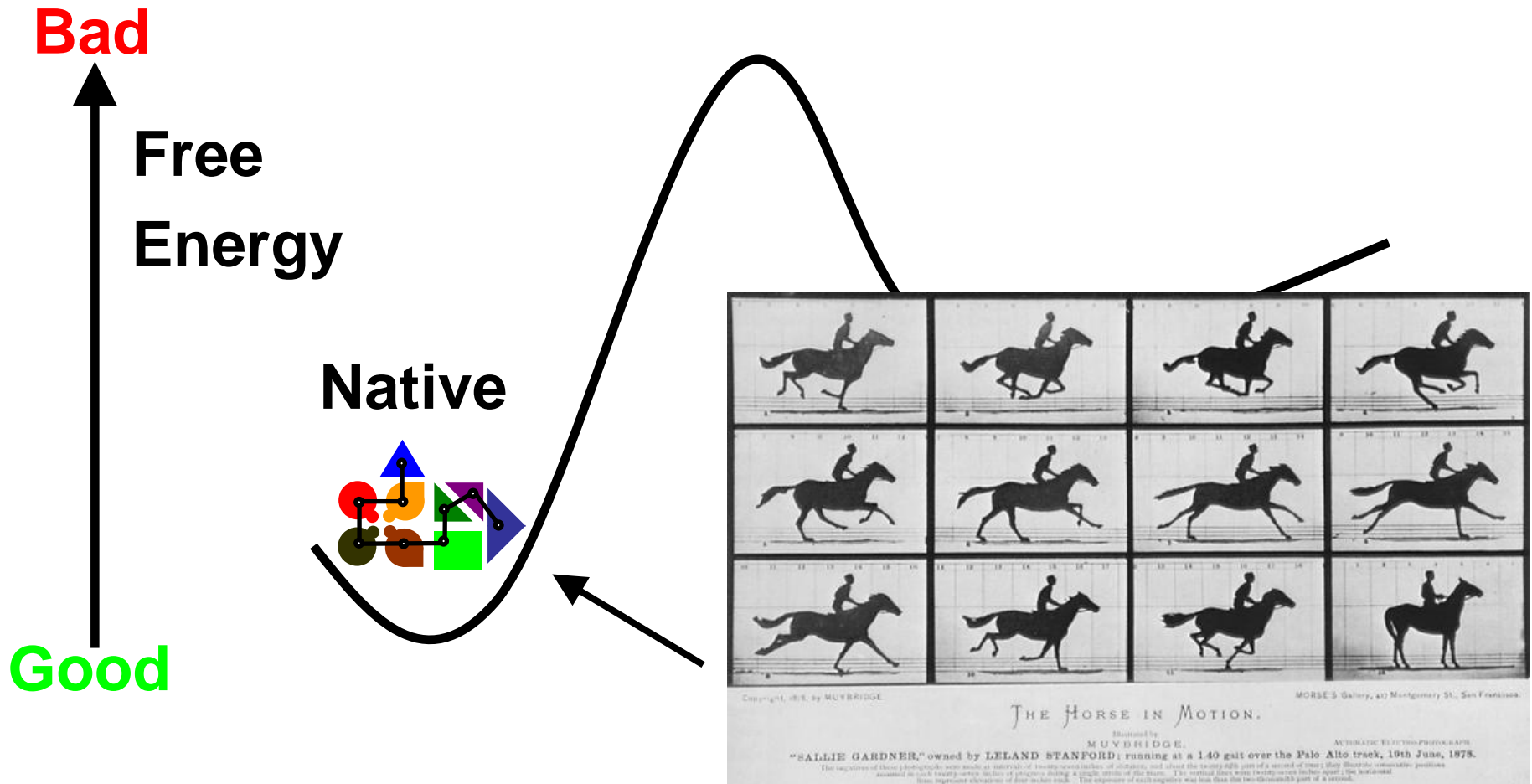
---

- Folded proteins are complex and dynamic molecules



# Proteins

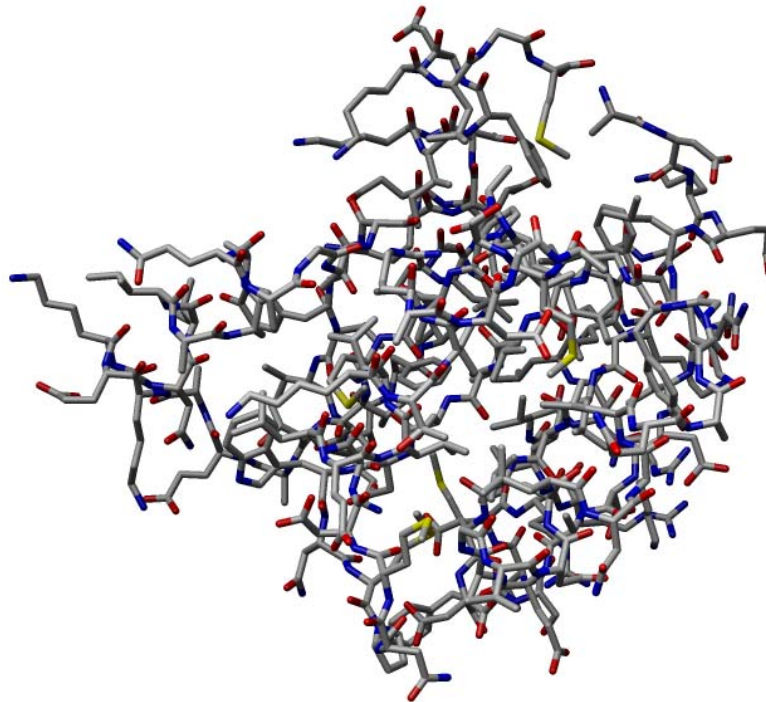
- Folded proteins are complex and dynamic molecules



# Molecular Dynamics

---

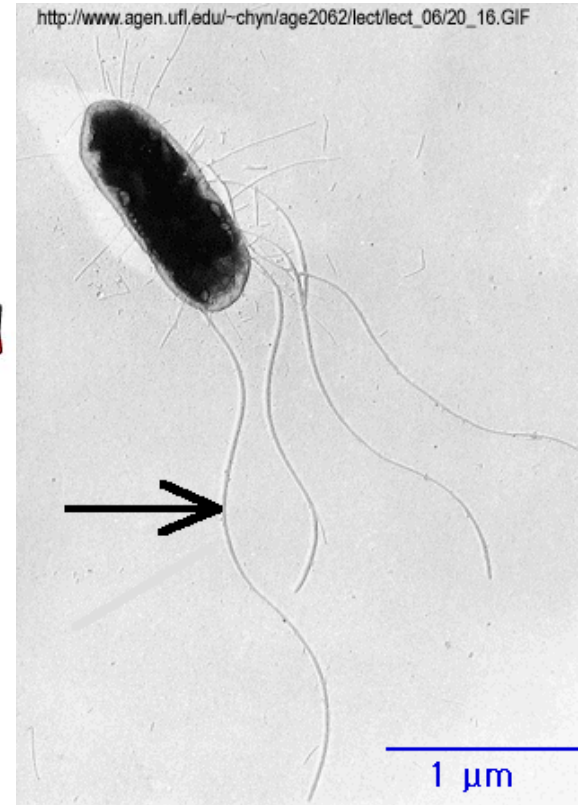
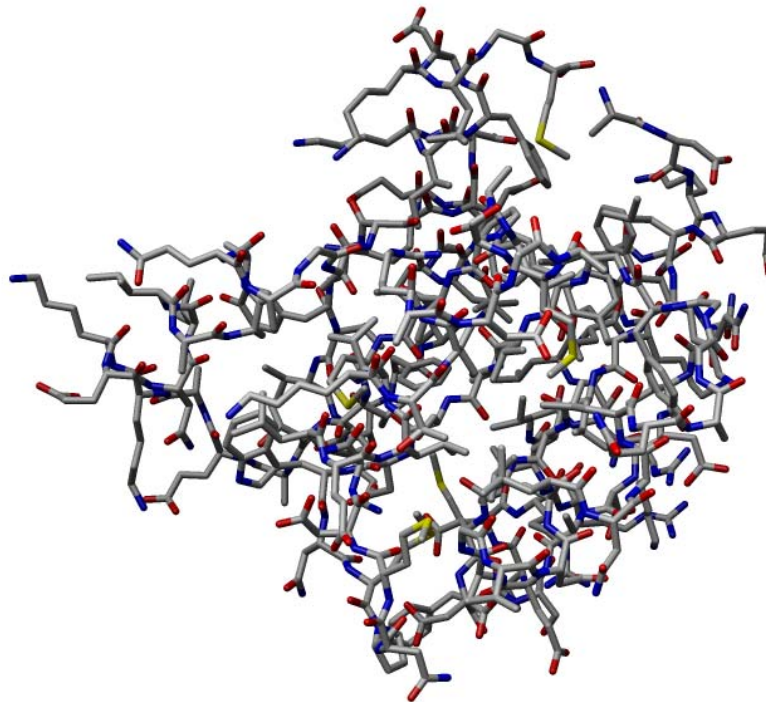
- MD provides atomic resolution of native dynamics



PDB ID: 3chy, *E. coli* CheY 1.66 Å X-ray crystallography

# Molecular Dynamics

- MD provides atomic resolution of native dynamics

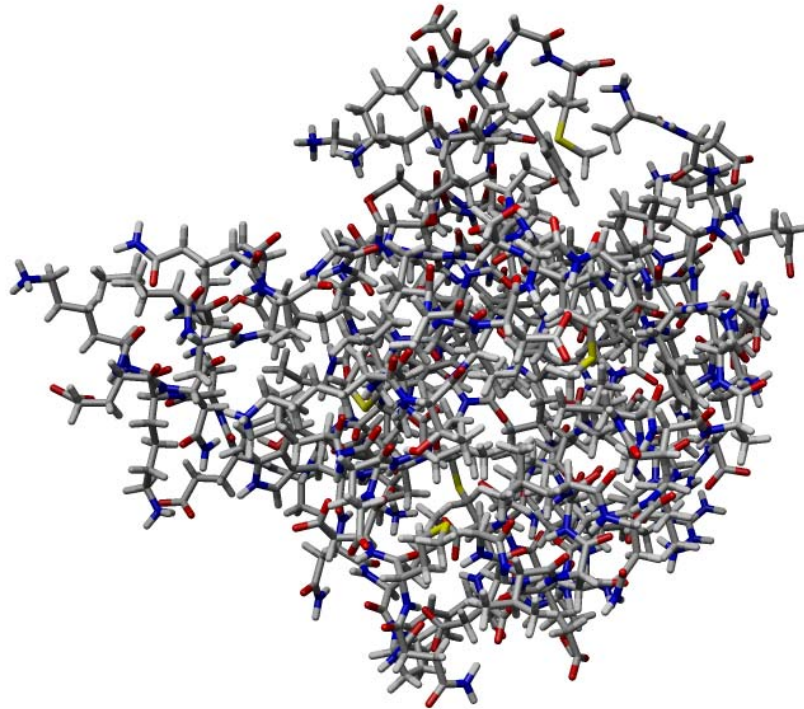


PDB ID: 3chy, *E. coli* CheY 1.66 Å X-ray crystallography

# Molecular Dynamics

---

- MD provides atomic resolution of native dynamics



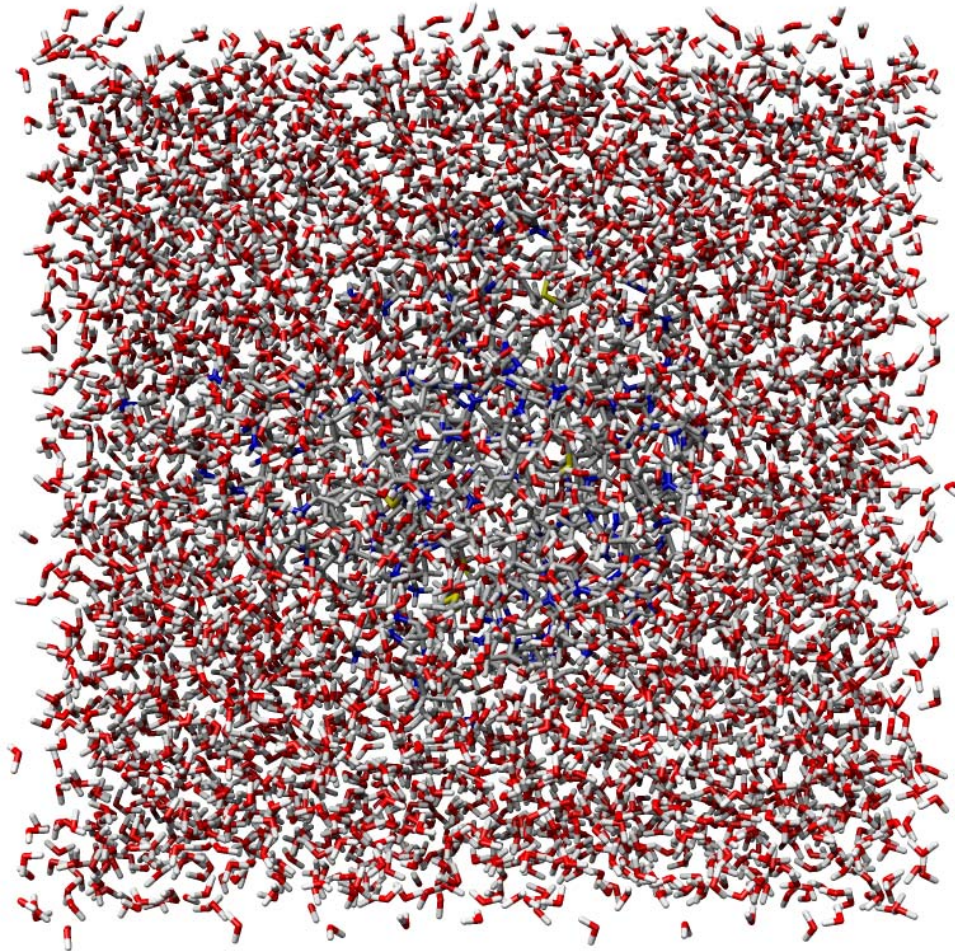
3chy, hydrogens added



# Molecular Dynamics

---

- MD provides atomic resolution of native dynamics



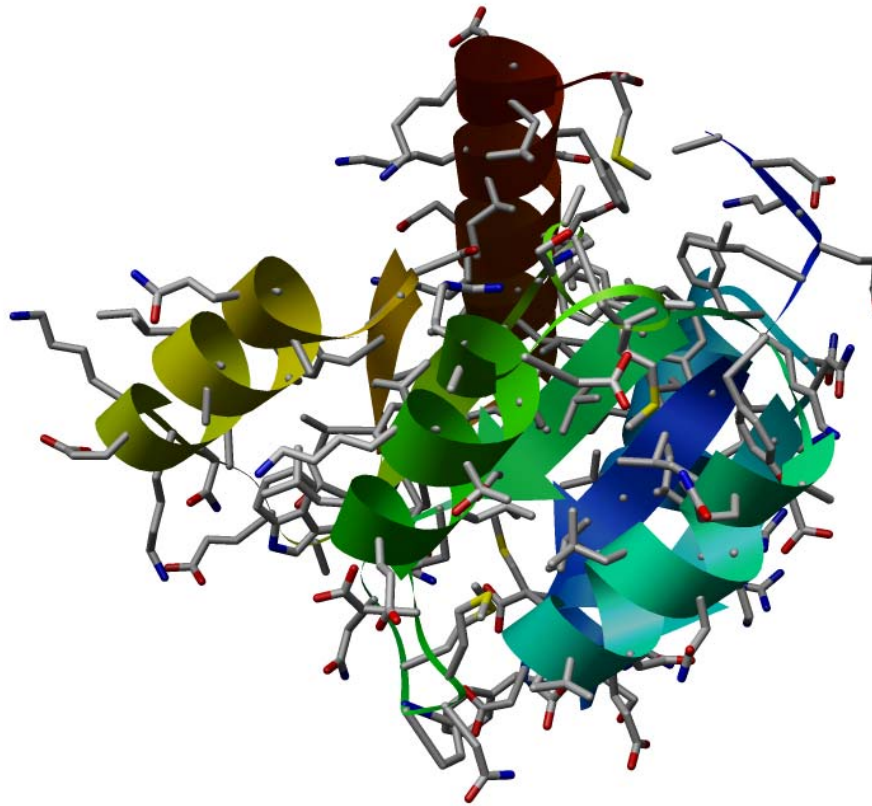
3chy, waters added (i.e. solvated)



# Molecular Dynamics

---

- MD provides atomic resolution of native dynamics

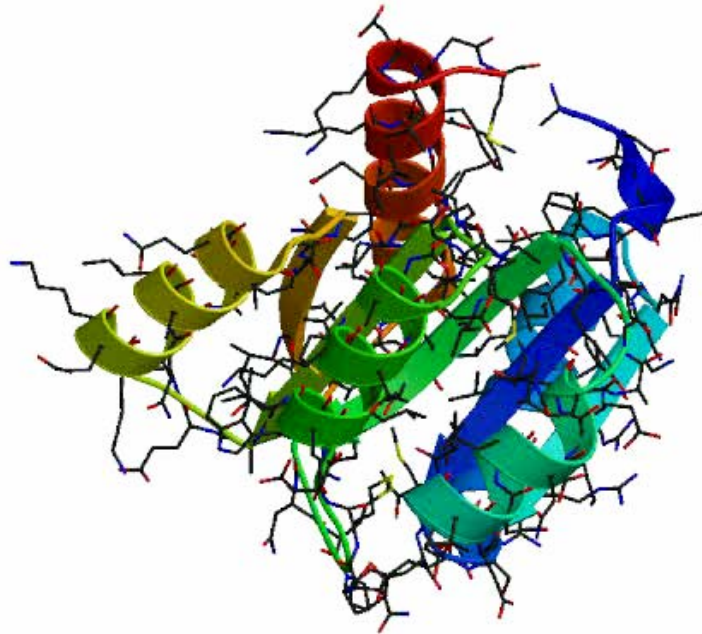


3chy, waters and hydrogens hidden

# Molecular Dynamics

---

- MD provides atomic resolution of native dynamics

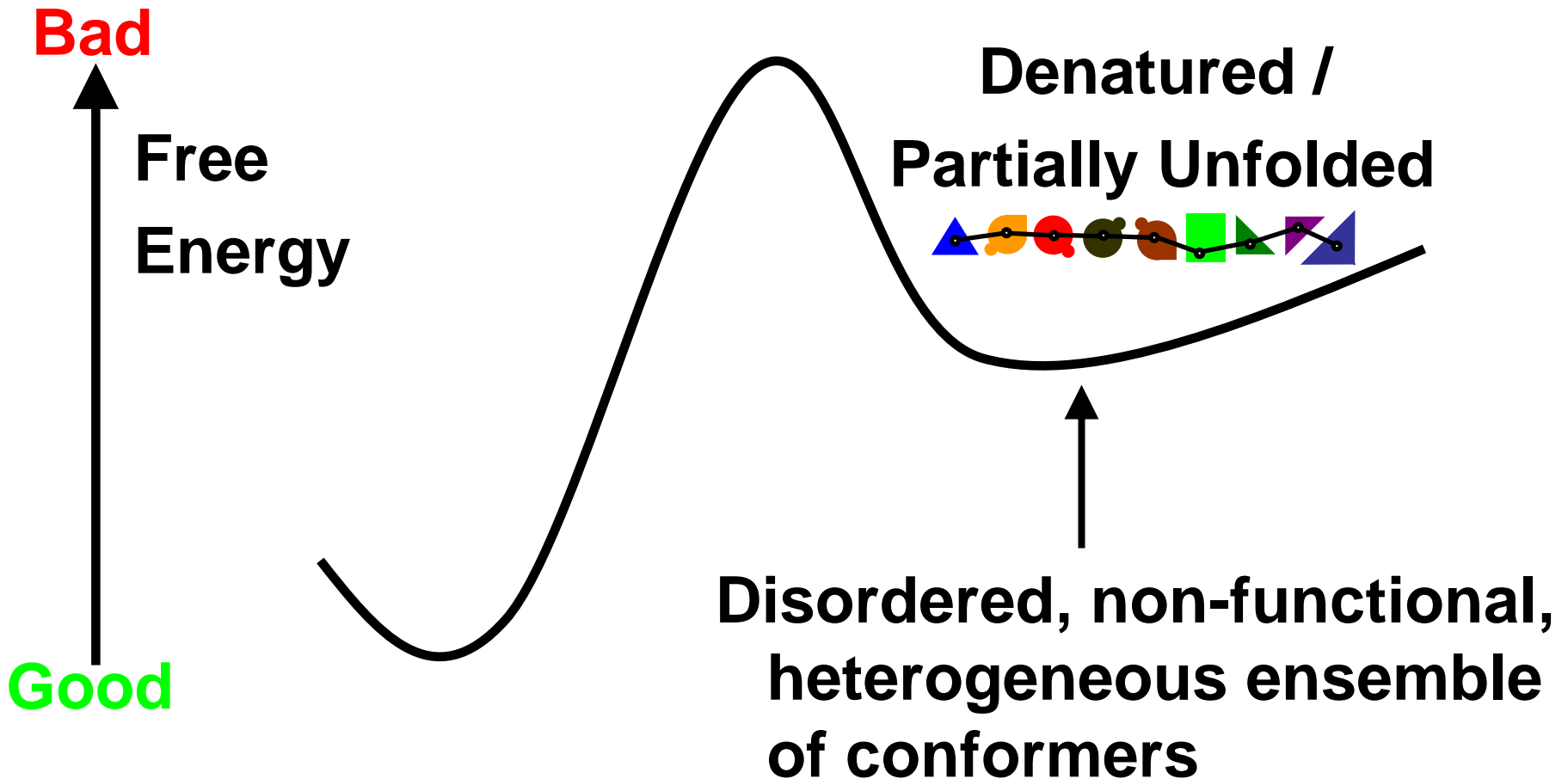


native state simulation of 3chy at 298 Kelvin, waters and hydrogens hidden

# Proteins

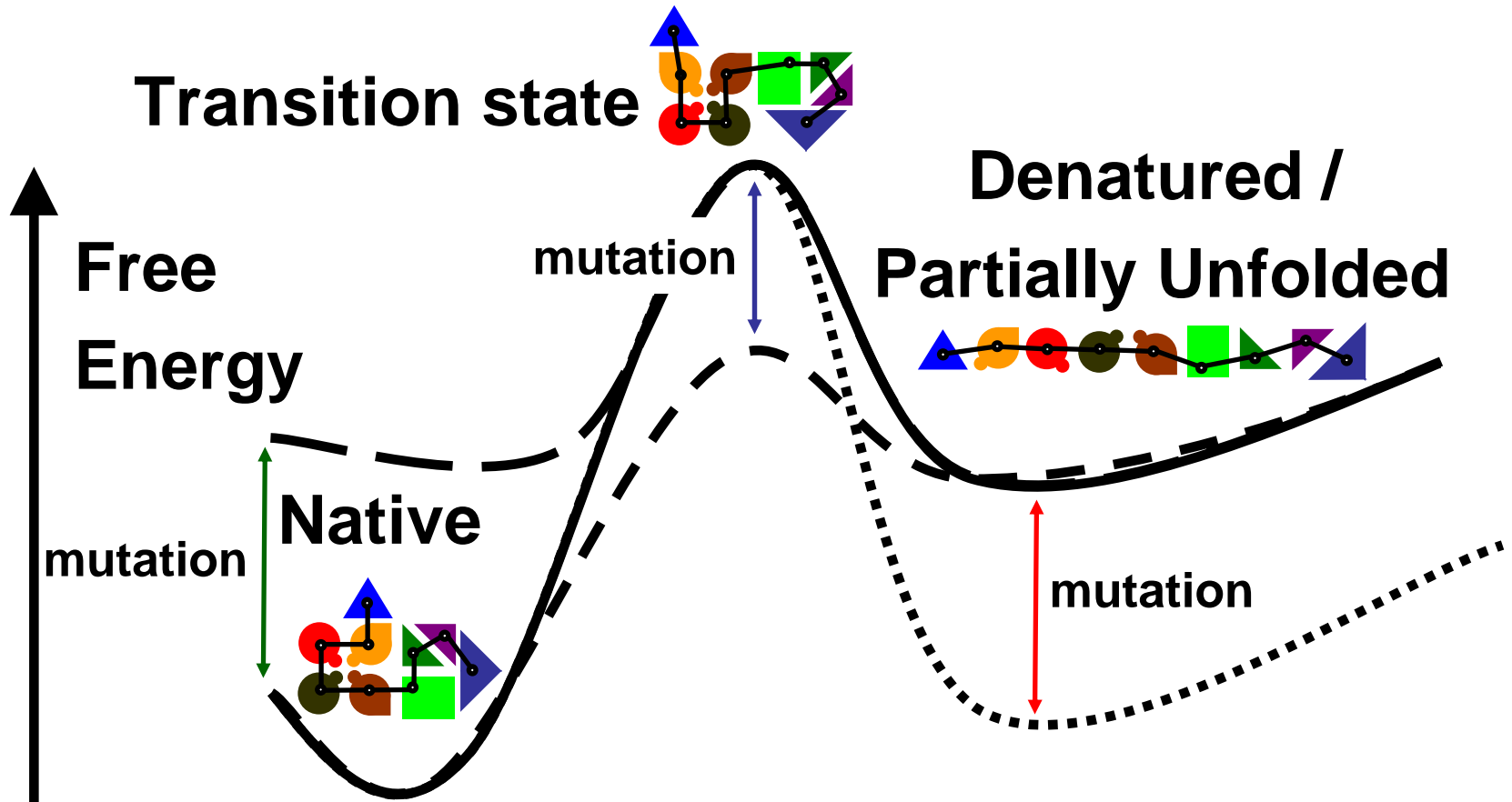
---

- Folding & unfolding at atomic resolution



# Proteins

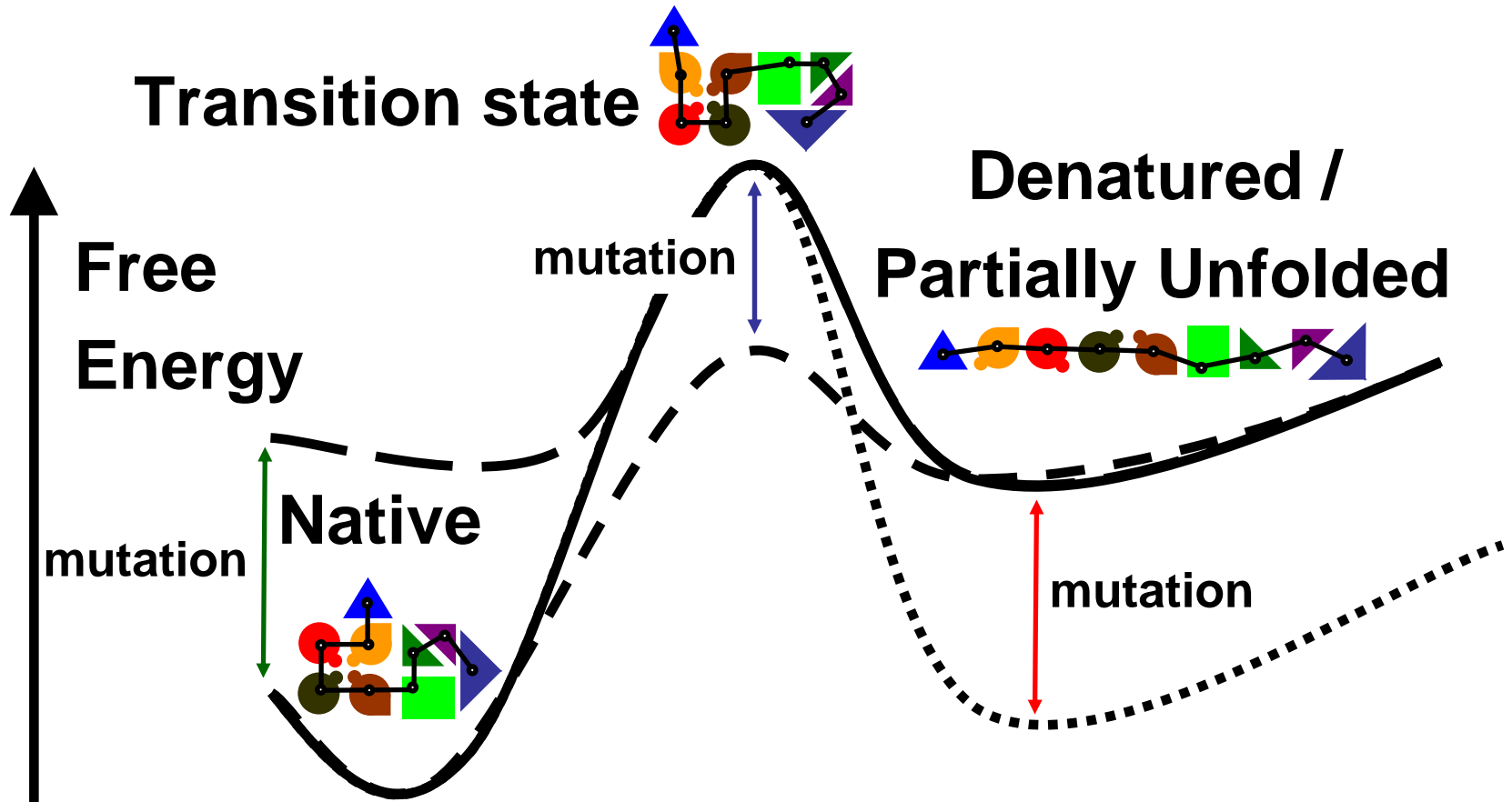
- Protein folding, why we care how it happens



Many diseases are related to protein folding and / or misfolding in response to genetic mutation.

# Proteins

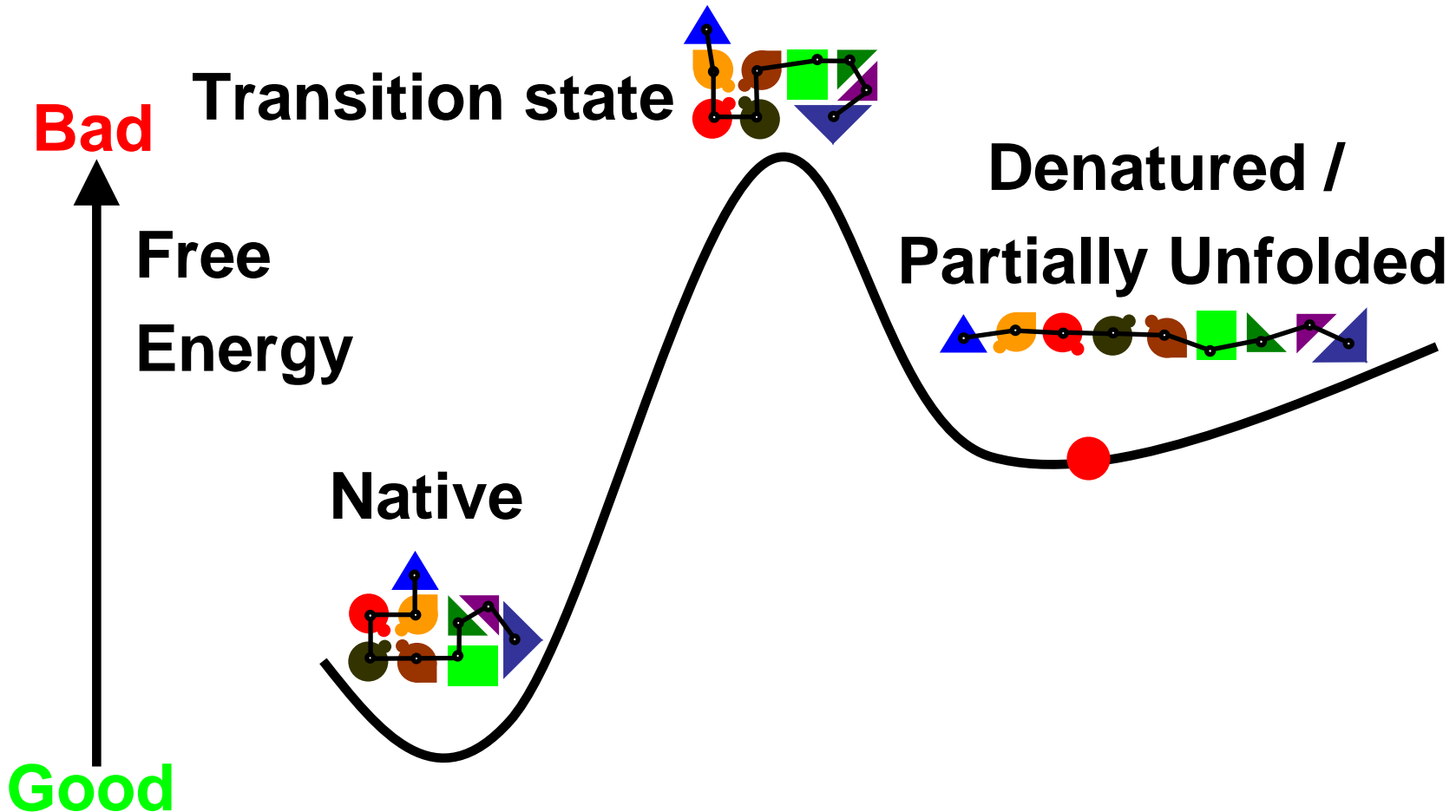
- Protein folding, why we care how it happens



We need to comprehend folding to build nano-scale biomachines (that could produce energy, etc...)

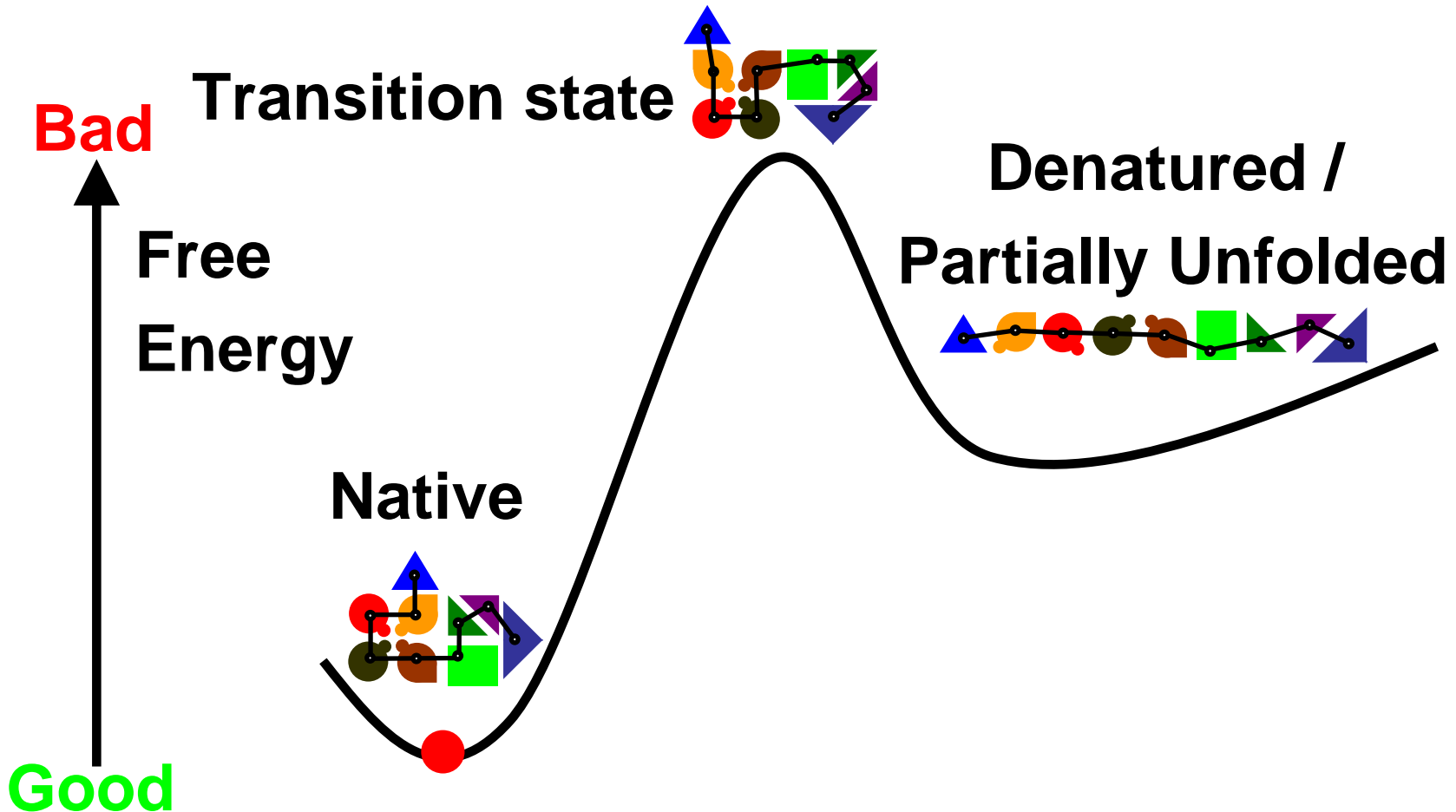
# Proteins

- Protein folding takes  $> 10 \mu\text{s}$  (often much longer)



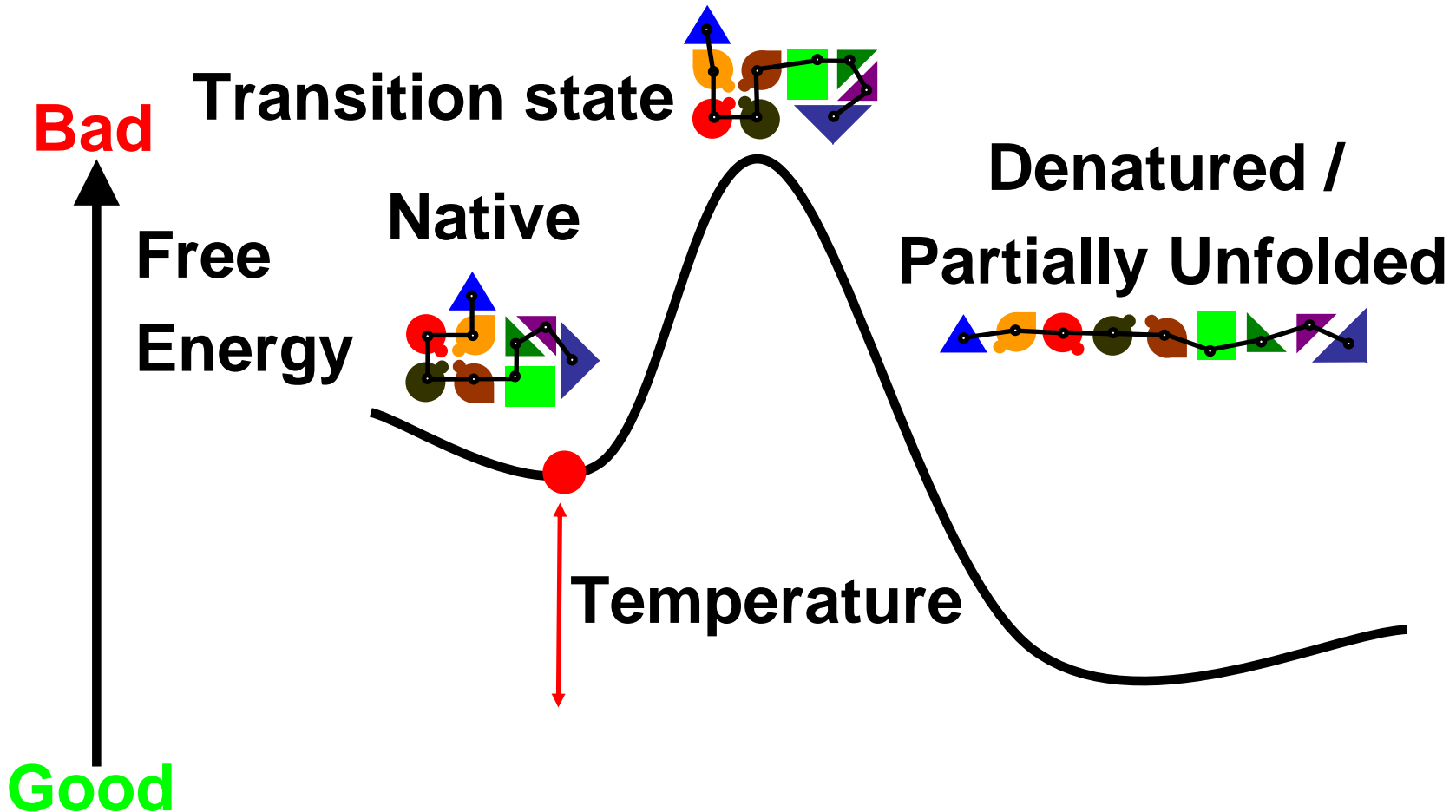
# Proteins

- Protein folding is the reverse of protein unfolding



# Proteins

- Protein unfolding is relatively invariant to temperature

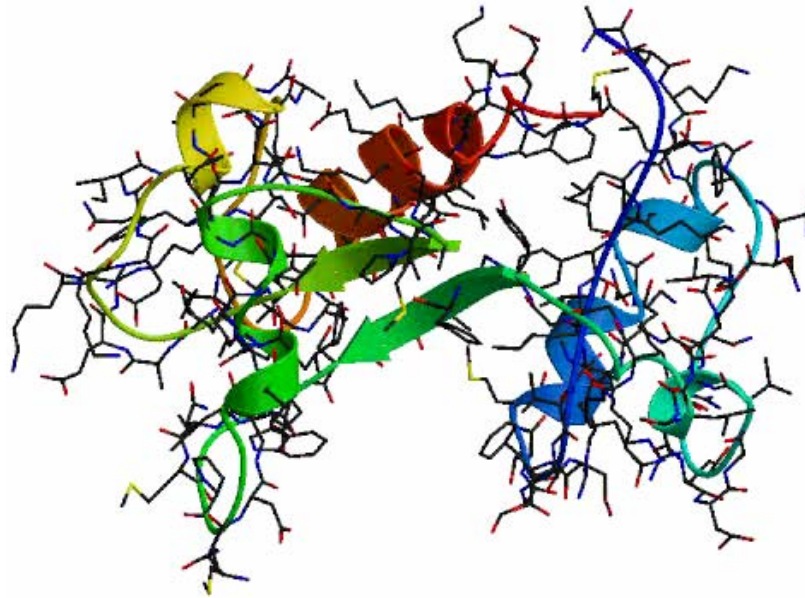




# Molecular Dynamics

---

- MD provides atomic resolution of folding / unfolding



unfolding simulation (reversed) of 3chy at 498 Kelvin, waters & hydrogens hidden

# Molecular Dynamics<sup>1</sup>

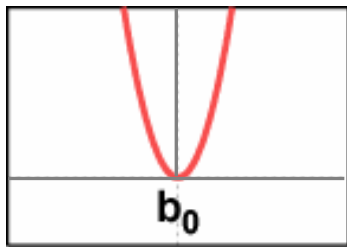
---

- Classically evolves an atomic system with time
  - **Potential function (a.k.a force field)**
    - Describes the energies of interaction between atom centers
  - **Integration algorithm**
    - Time dependent evolution of atomic coordinates in response to potential energy
  - **Statistical sampling ensemble**
    - Fixed thermodynamic variables, i.e. NVE
    - Number of atoms, box Volume, total Energy

# Molecular Dynamics

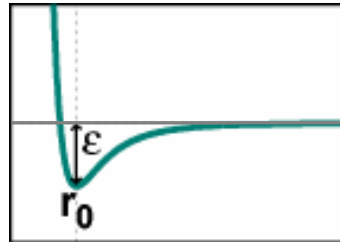
- Potential function for MD<sup>1,2</sup>

$$U = \text{Bond} + \text{Angle} + \text{Dihedral} + \text{van der Waals} + \text{Electrostatic}$$



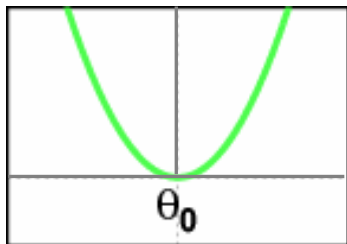
**Bond**

$$\sum_i^{\text{bonds}} K_{b,i} (b_i - b_{0,i})^2$$



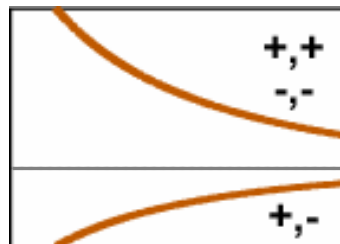
**van der Waals**

$$\sum_{\text{pairs } i,j} \left[ \epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^{12} - 2\epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^6 \right]$$



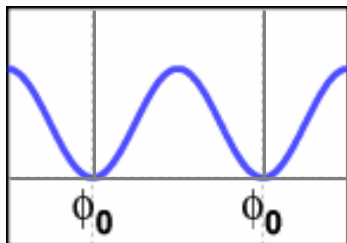
**Angle**

$$\sum_i^{\text{bond angles}} K_{\theta,i} (\theta_i - \theta_{0,i})^2$$



**Electrostatic**

$$332 \sum_{\text{pairs } i,j} \left( \frac{q_i q_j}{r_{ij}} \right)$$



**Dihedral**

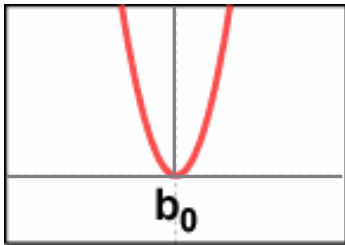
$$\sum_i^{\text{torsion angles}} K_{\phi,i} \{1 - \cos[n_i (\phi_i - \phi_{0,i})]\}$$

- Levitt M. Hirshberg M. Sharon R. Daggett V. Comp. Phys. Comm. (1995) 91: 215-231
- Levitt M. *et al.* J. Phys. Chem. B (1997) 101: 5051-5061

# Molecular Dynamics

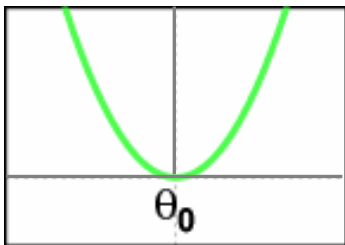
- Potential function for MD<sup>1,2</sup>

$U = \text{Bond} + \text{Angle} + \text{Dihedral} + \text{van der Waals} + \text{Electrostatic}$



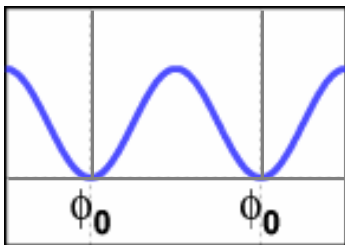
**Bond**

$$\sum_i^{\text{bonds}} K_{b,i} (b_i - b_{0,i})^2$$



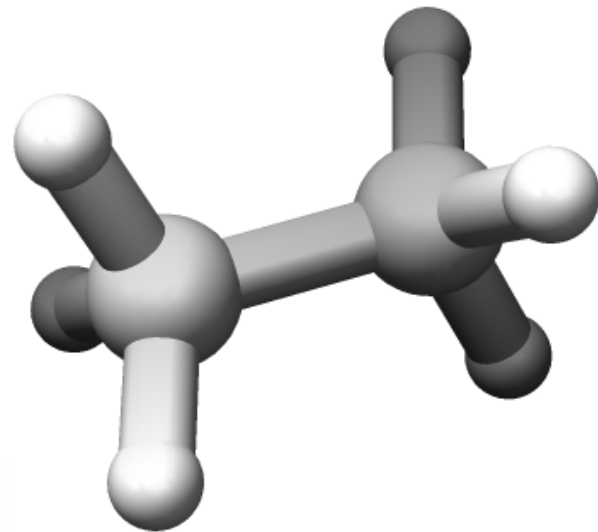
**Angle**

$$\sum_i^{\text{bond angles}} K_{\theta,i} (\theta_i - \theta_{0,i})^2$$



**Dihedral**

$$\sum_i^{\text{torsion angles}} K_{\phi,i} \{1 - \cos[n_i (\phi_i - \phi_{0,i})]\}$$

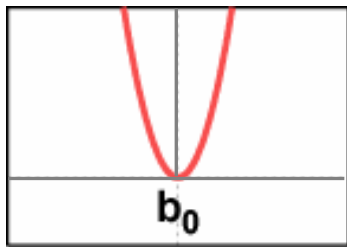


1. Levitt M. Hirshberg M. Sharon R. Daggett V. Comp. Phys. Comm. (1995) 91: 215-231
2. Levitt M. *et al.* J. Phys. Chem. B (1997) 101: 5051-5061

# Molecular Dynamics

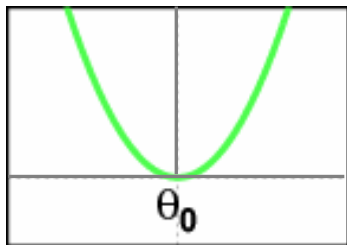
- Potential function for MD<sup>1,2</sup>

$U = \text{Bond} + \text{Angle} + \text{Dihedral} + \text{van der Waals} + \text{Electrostatic}$



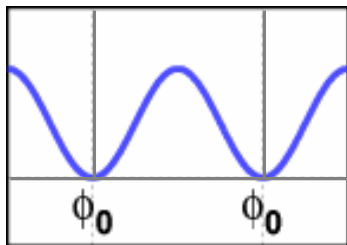
**Bond**

$$\sum_i^{\text{bonds}} K_{b,i} (b_i - b_{0,i})^2$$



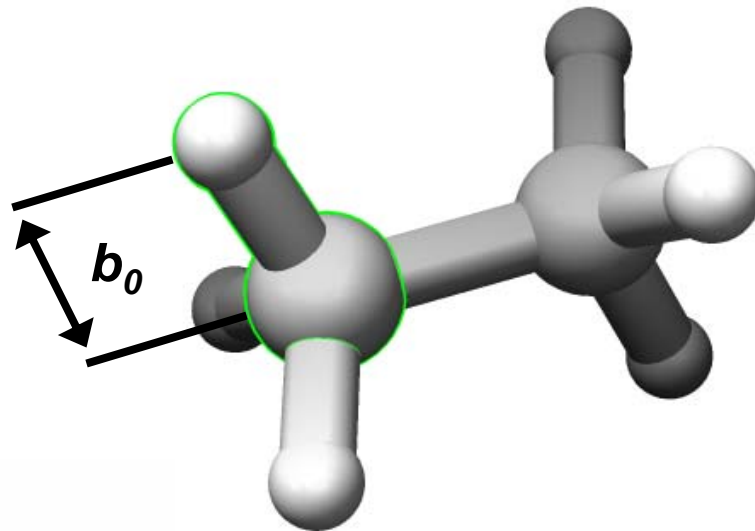
**Angle**

$$\sum_i^{\text{bond angles}} K_{\theta,i} (\theta_i - \theta_{0,i})^2$$



**Dihedral**

$$\sum_i^{\text{torsion angles}} K_{\phi,i} \{1 - \cos[n_i (\phi_i - \phi_{0,i})]\}$$

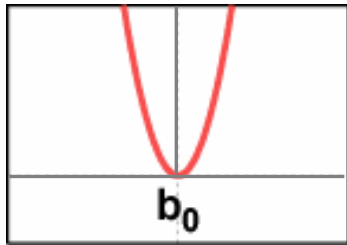


1. Levitt M. Hirshberg M. Sharon R. Daggett V. Comp. Phys. Comm. (1995) 91: 215-231
2. Levitt M. *et al.* J. Phys. Chem. B (1997) 101: 5051-5061

# Molecular Dynamics

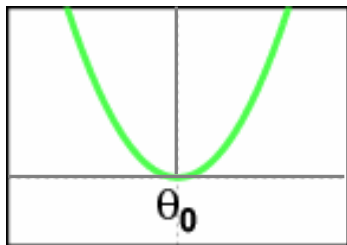
- Potential function for MD<sup>1,2</sup>

$U = \text{Bond} + \text{Angle} + \text{Dihedral} + \text{van der Waals} + \text{Electrostatic}$



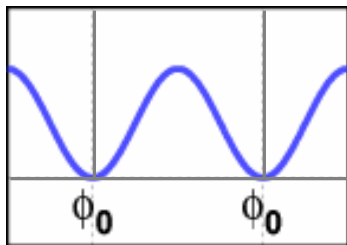
**Bond**

$$\sum_i^{\text{bonds}} K_{b,i} (b_i - b_{0,i})^2$$



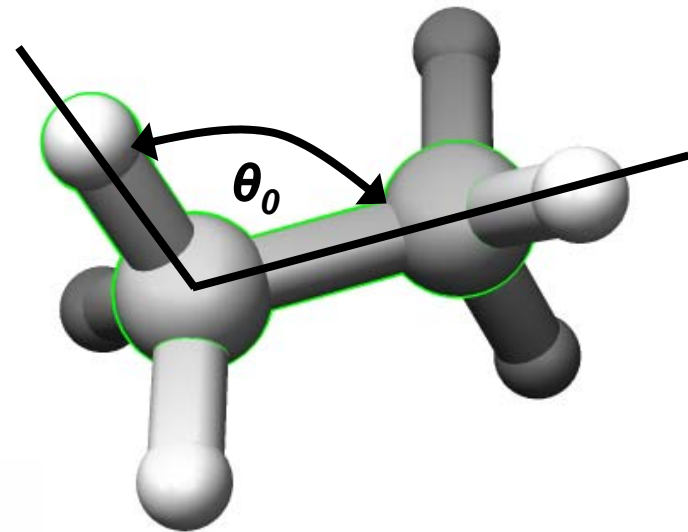
**Angle**

$$\sum_i^{\text{bond angles}} K_{\theta,i} (\theta_i - \theta_{0,i})^2$$



**Dihedral**

$$\sum_i^{\text{torsion angles}} K_{\phi,i} \{1 - \cos[n_i (\phi_i - \phi_{0,i})]\}$$

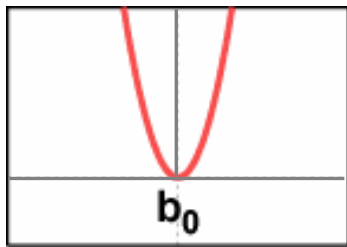


1. Levitt M. Hirshberg M. Sharon R. Daggett V. Comp. Phys. Comm. (1995) 91: 215-231
2. Levitt M. *et al.* J. Phys. Chem. B (1997) 101: 5051-5061

# Molecular Dynamics

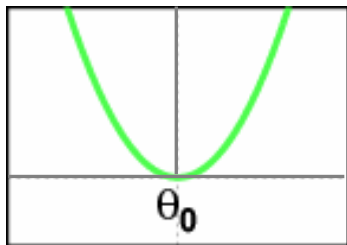
- Potential function for MD<sup>1,2</sup>

$$U = \text{Bond} + \text{Angle} + \text{Dihedral} + \text{van der Waals} + \text{Electrostatic}$$



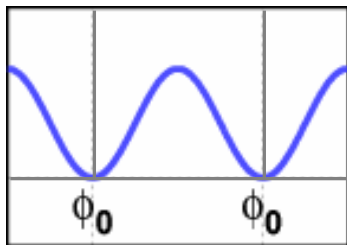
**Bond**

$$\sum_i^{\text{bonds}} K_{b,i} (b_i - b_{0,i})^2$$



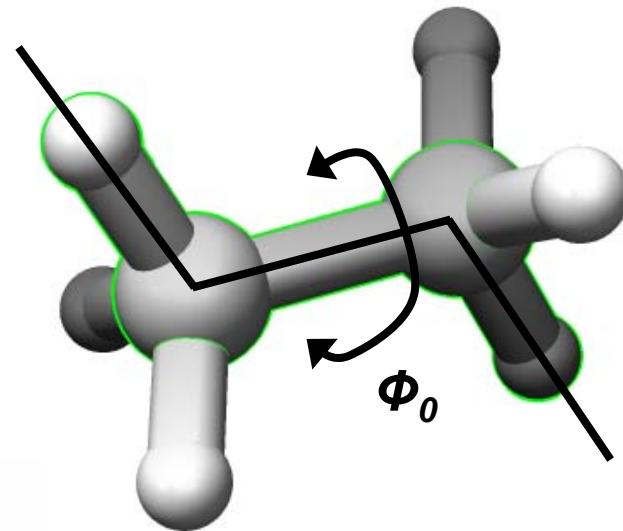
**Angle**

$$\sum_i^{\text{bond angles}} K_{\theta,i} (\theta_i - \theta_{0,i})^2$$



**Dihedral**

$$\sum_i^{\text{torsion angles}} K_{\phi,i} \{1 - \cos[n_i (\phi_i - \phi_{0,i})]\}$$

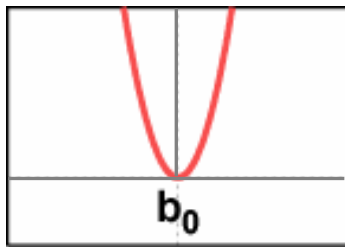


- Levitt M. Hirshberg M. Sharon R. Daggett V. Comp. Phys. Comm. (1995) 91 215-231
- Levitt M. *et al.* J. Phys. Chem. B (1997) 101:25 5051-5061

# Molecular Dynamics

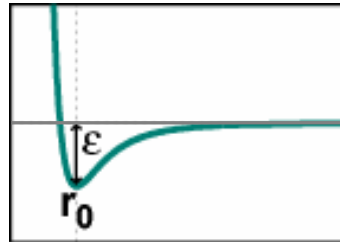
- Potential function for MD<sup>1,2</sup>

$$U = \text{Bond} + \text{Angle} + \text{Dihedral} + \text{van der Waals} + \text{Electrostatic}$$



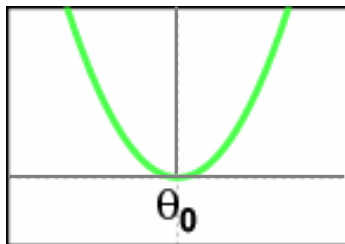
**Bond**

$$\sum_i^{\text{bonds}} K_{b,i} (b_i - b_{0,i})^2$$



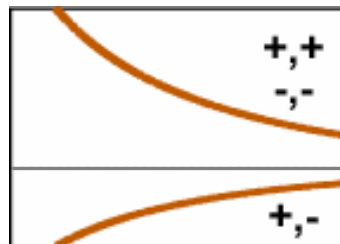
**van der Waals**

$$\sum_{\text{pairs } i,j} \left[ \epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^{12} - 2\epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^6 \right]$$



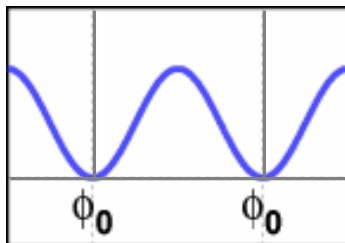
**Angle**

$$\sum_i^{\text{bond angles}} K_{\theta,i} (\theta_i - \theta_{0,i})^2$$



**Electrostatic**

$$332 \sum_{\text{pairs } i,j} \left( \frac{q_i q_j}{r_{ij}} \right)$$



**Dihedral**

$$\sum_i^{\text{torsion angles}} K_{\phi,i} \{1 - \cos[n_i (\phi_i - \phi_{0,i})]\}$$

- Levitt M. Hirshberg M. Sharon R. Daggett V. Comp. Phys. Comm. (1995) 91: 215-231
- Levitt M. *et al.* J. Phys. Chem. B (1997) 101: 5051-5061

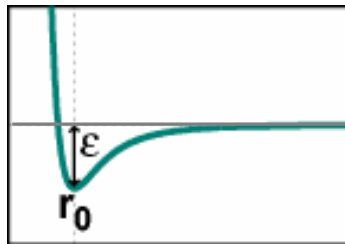


# Molecular Dynamics

---

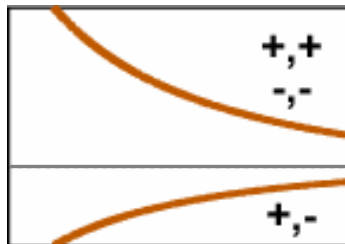
- Non-bonded components of potential function

$$U_{nb} = \text{van der Waals} + \text{Electrostatic}$$



van der Waals

$$\sum_{\text{pairs } i,j} \left[ \epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^{12} - 2\epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^6 \right]$$



Electrostatic

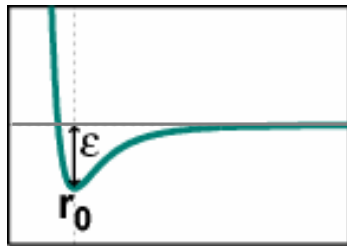
$$\frac{1}{4\pi\epsilon_0} \sum_{\text{pairs } i,j} \left( \frac{q_i q_j}{r_{ij}} \right)$$

- *To a large degree, protein structure is dependent on non-bonded atomic interactions*

# Molecular Dynamics

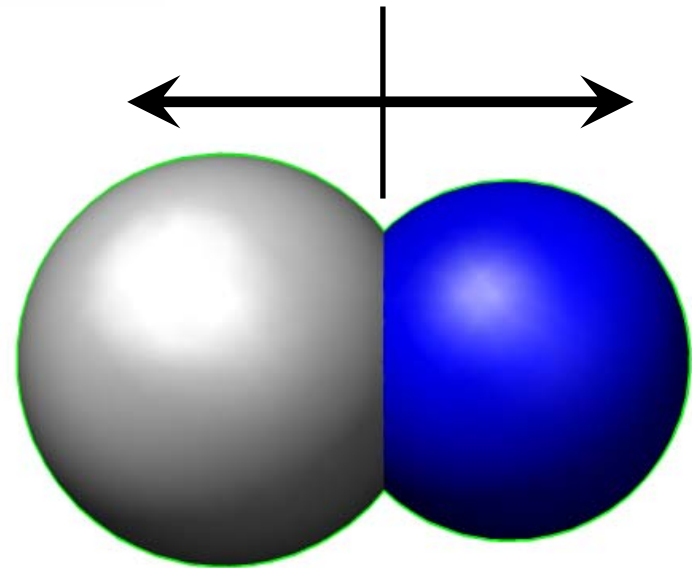
- Non-bonded components of potential function

$U_{nb}$  = van der Waals + Electrostatic



van der Waals

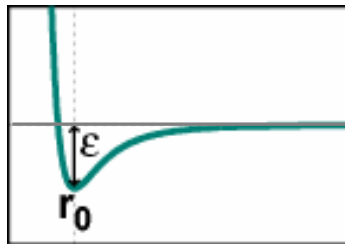
$$\sum_{\text{pairs } i,j} \left[ \epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^{12} - 2\epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^6 \right]$$



# Molecular Dynamics

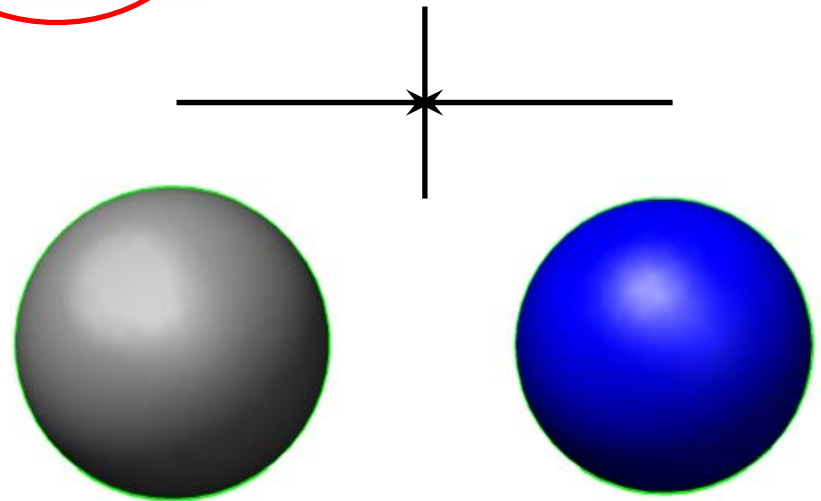
- Non-bonded components of potential function

$U_{nb}$  = van der Waals + Electrostatic



van der Waals

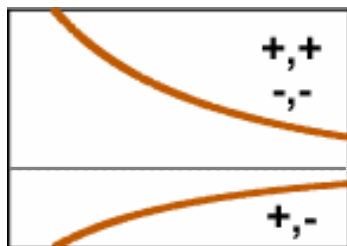
$$\sum_{\text{pairs } i,j} \left[ \epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^{12} - 2\epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^6 \right]$$



# Molecular Dynamics

- Non-bonded components of potential function

$$U_{nb} = \text{van der Waals} + \text{Electrostatic}$$



Electrostatic

$$\sum_{\text{pairs } i,j} \left( \frac{q_i q_j}{r_{ij}} \right)$$



+

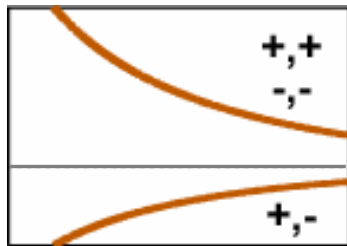


-

# Molecular Dynamics

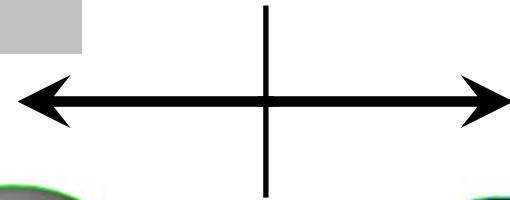
- Non-bonded components of potential function

$$U_{nb} = \text{van der Waals} + \text{Electrostatic}$$



Electrostatic

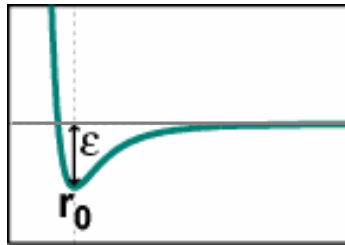
$$\sum_{\text{pairs } i,j} \left( \frac{q_i q_j}{r_{ij}} \right)$$



# Molecular Dynamics

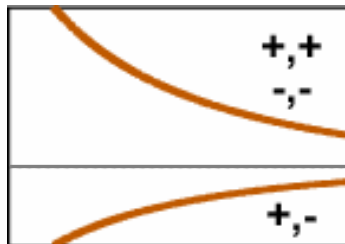
- Non-bonded components of potential function

$$U_{nb} = \text{van der Waals} + \text{Electrostatic}$$



van der Waals

$$\sum_{\text{pairs } i,j} \left[ \epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^{12} - 2\epsilon_{ij} \left( \frac{r_{0,ij}}{r_{ij}} \right)^6 \right]$$



Electrostatic

$$\sum_{\text{pairs } i,j} \left( \frac{q_i q_j}{r_{ij}} \right)$$

**NOTE:**

Sum over all pairs of N atoms, or

$$\frac{N * N - 1}{2} \text{ pairs}$$

N is often between  $5 \times 10^5$  to  $5 \times 10^6$

For  $5 \times 10^5$  that is  $1.25 \times 10^{11}$  pairs

**THAT IS A LOT OF POSSIBLE PAIRS!**

# Molecular Dynamics

---

- Time dependent integration of classical equations of motion

$$F = -\frac{\partial U}{\partial x}$$

$$F = ma$$

$$a = \frac{v_2 - v_1}{\partial t}$$

$$v = \frac{x_2 - x_1}{\partial t}$$

$$E = U + K$$

$$\partial t = 2 \text{ fs}$$

# Molecular Dynamics

---

- Time dependent integration

$$F = -\frac{\partial U}{\partial x}$$

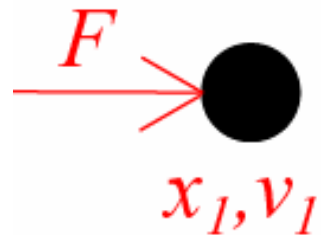
$$F = ma$$

$$a = \frac{v_2 - v_1}{\partial t}$$

$$v = \frac{x_2 - x_1}{\partial t}$$

$$E = U + K$$

$$\partial t = 2 \text{ fs}$$





# Molecular Dynamics

---

- Time dependent integration

$$F = -\frac{\partial U}{\partial x}$$

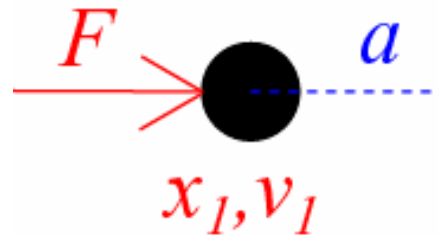
$$F = ma$$

$$a = \frac{v_2 - v_1}{\partial t}$$

$$v = \frac{x_2 - x_1}{\partial t}$$

$$E = U + K$$

$$\partial t = 2 \text{ fs}$$



# Molecular Dynamics

---

- Time dependent integration

$$F = -\frac{\partial U}{\partial x}$$

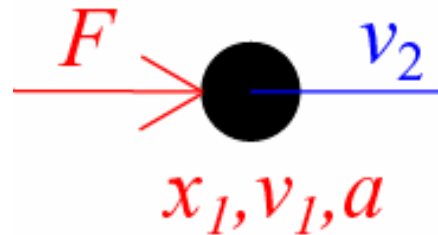
$$F = ma$$

$$a = \frac{v_2 - v_1}{\partial t}$$

$$v = \frac{x_2 - x_1}{\partial t}$$

$$E = U + K$$

$$\partial t = 2 \text{ fs}$$



# Molecular Dynamics

---

- Time dependent integration

$$F = -\frac{\partial U}{\partial x}$$

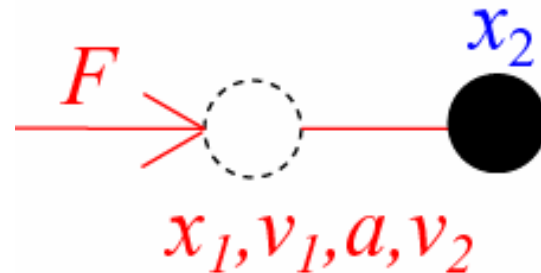
$$F = ma$$

$$a = \frac{v_2 - v_1}{\partial t}$$

$$v = \frac{x_2 - x_1}{\partial t}$$

$$E = U + K$$

$$\partial t = 2 \text{ fs}$$



# Molecular Dynamics

---

- Time dependent integration

$$F = -\frac{\partial U}{\partial x}$$

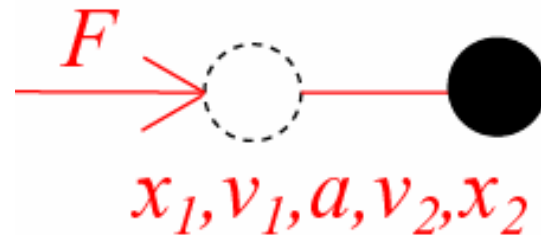
$$F = ma$$

$$a = \frac{v_2 - v_1}{\partial t}$$

$$v = \frac{x_2 - x_1}{\partial t}$$

$$E = U + K$$

$$\partial t = 2 \text{ fs}$$



# Molecular Dynamics

---

- Time dependent integration

$$F = -\frac{\partial U}{\partial x}$$

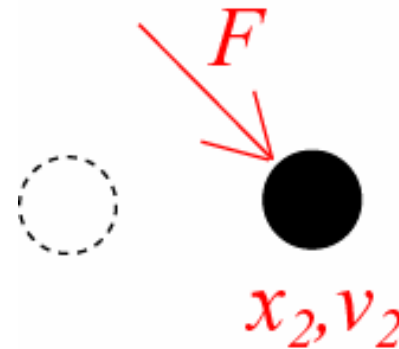
$$F = ma$$

$$a = \frac{v_3 - v_2}{\partial t}$$

$$v = \frac{x_3 - x_2}{\partial t}$$

$$E = U + K$$

$$\partial t = 2 \text{ fs}$$



# Molecular Dynamics

---

- Time dependent integration

$$F = -\frac{\partial U}{\partial x}$$

$$F = ma$$

$$a = \frac{v_3 - v_2}{\partial t}$$

$$v = \frac{x_3 - x_2}{\partial t}$$

$$E = U + K$$

$$\partial t = 2 \text{ fs}$$

**Evaluate forces and  
perform integration for  
every atom**

**Each picosecond of  
simulation time requires  
500 iterations of cycle**

**E.g. w/ 50,000 atoms, each  
ps ( $10^{-12}$  s) involves  
25,000,000 evaluations**

# Molecular Dynamics

---

- Scalable, parallel MD & analysis software:

***ihmm***

***in lucem* Molecular Mechanics<sup>1</sup>**

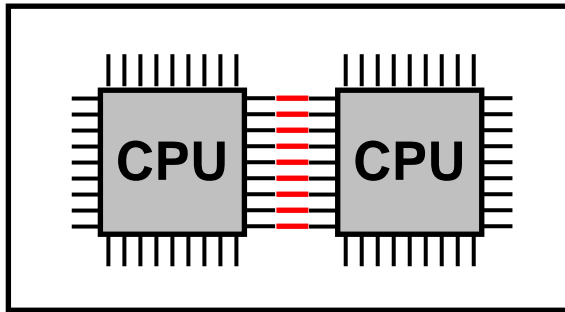
1. Beck, Alonso, Daggett, (2004) University of Washington, Seattle

# Molecular Dynamics

---

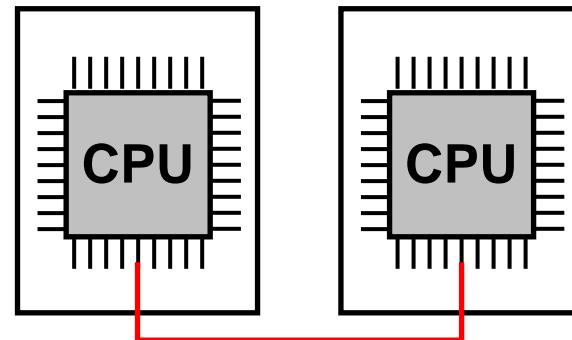
- *lmm* is written in C (ANSI / POSIX)
- 64 bit math
- POSIX threads / MPI

POSIX threads  
(multiprocessor machines)



+

Message Passing Interface  
(multiple machines)



**VERY** high bandwidth

- **Software design philosophy:**
  - Kernel
    - Compiles user's molecular mechanics programs
    - Schedules execution across processor and machines
  - Modules, e.g.
    - Molecular Dynamics
    - Analysis

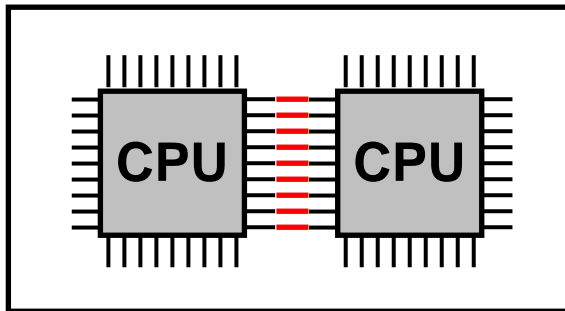


# Molecular Dynamics

---

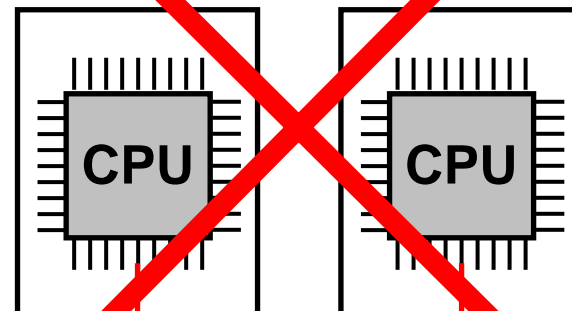
- *lmm* is written in C (ANSI / POSIX)
- 64 bit math
- POSIX threads / MPI

POSIX threads  
(multiprocessor machines)



+

Message Passing Interface  
(multiple machines)



- **Software design philosophy:**

- Kernel

- Compiles user's molecular mechanics programs
- Schedules execution across processor and machines

- Modules, e.g.

- Molecular Dynamics
- Analysis

~~VERY high bandwidth~~

# Dynameomics

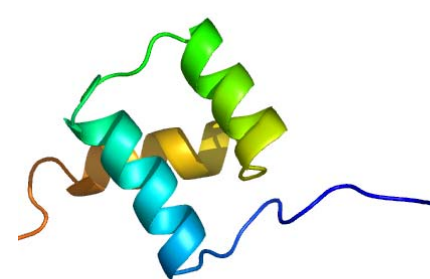
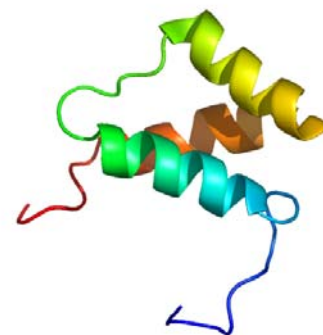
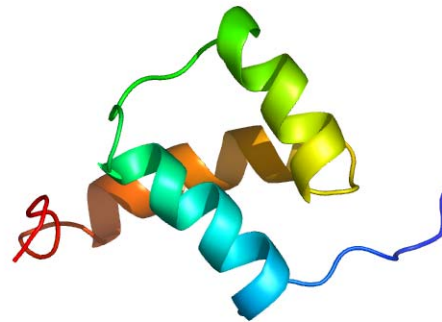
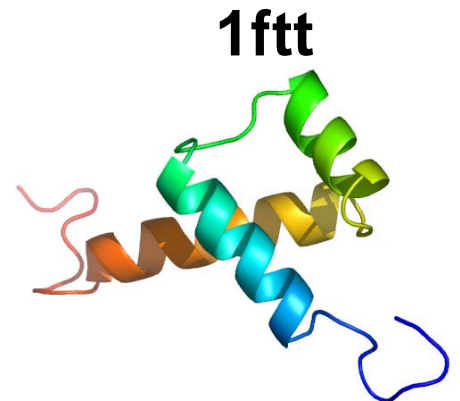
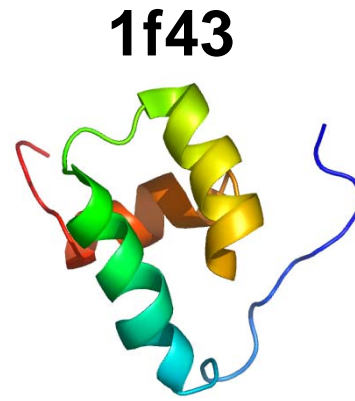
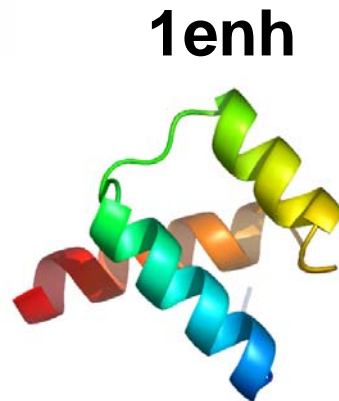
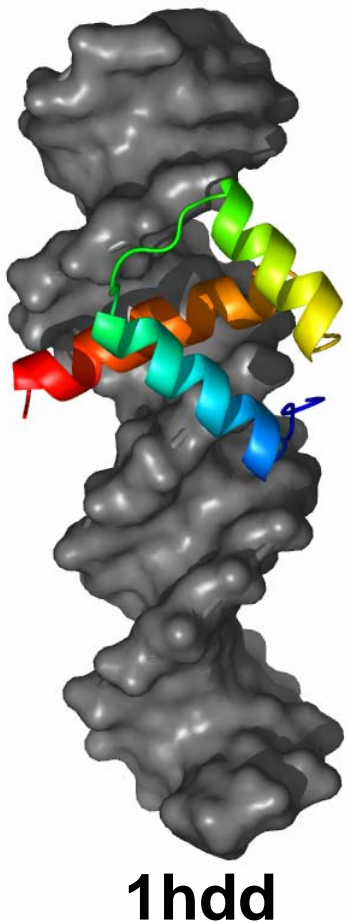
---

- **Simulate representative protein from all folds**

# Dynameomics

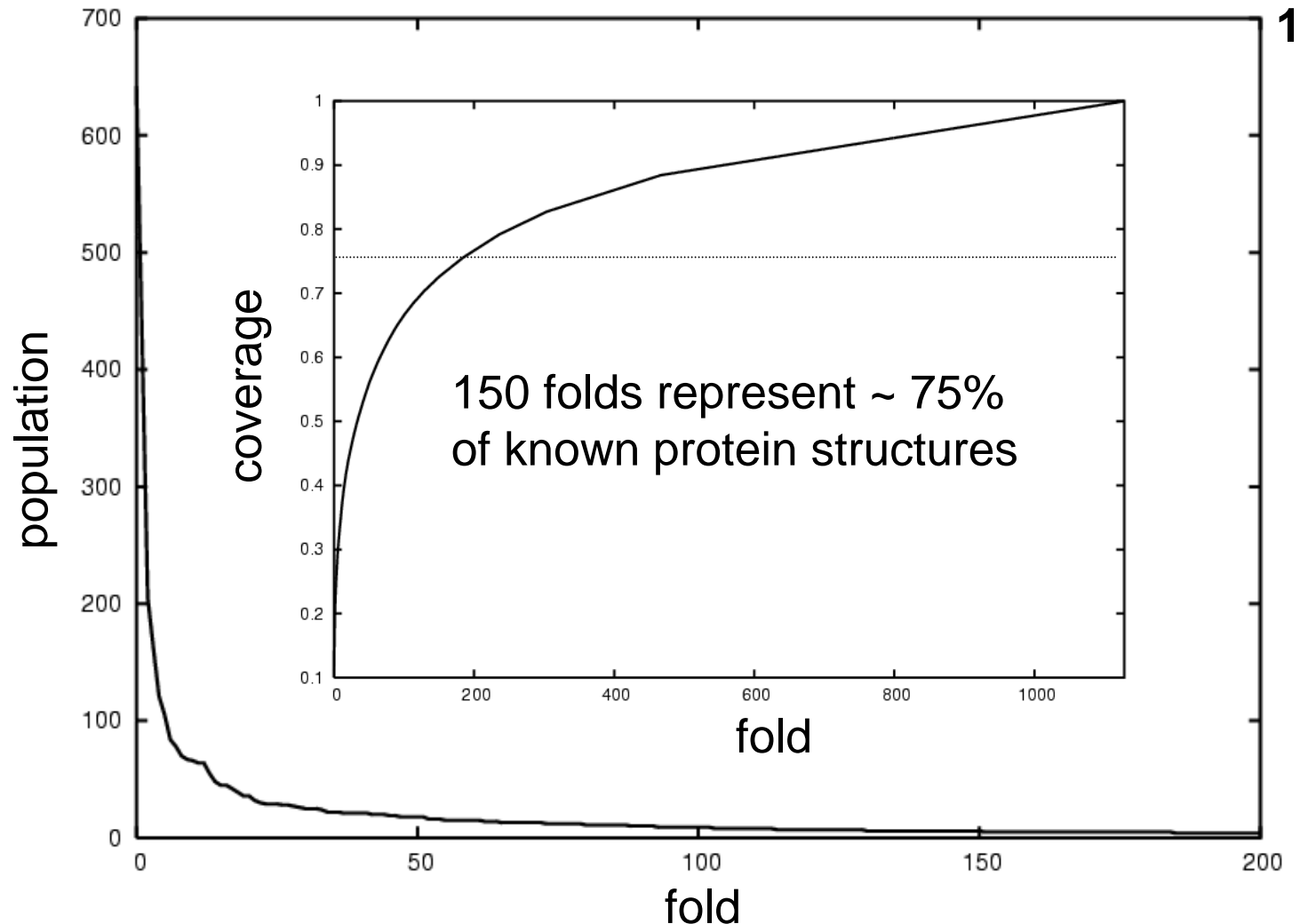
---

- **Simulate representative protein from all folds**
  - Nature reuses designs for similar jobs



# Dynameomics

- Simulate representative protein from all folds



# Dynameomics

---

- **Simulate representative protein from all folds**
  - **Native (folded) dynamics**
    - 20 nanosecond simulation at 298 Kelvin
  - **Folding / unfolding pathway**
    - 3 x 2 ns simulations at 498 K
    - 2 x 20 ns simulations at 498 K
  - **Each target requires 6 simulations**

**=**

**MANY CPU HOURS**

# Dynameomics

---

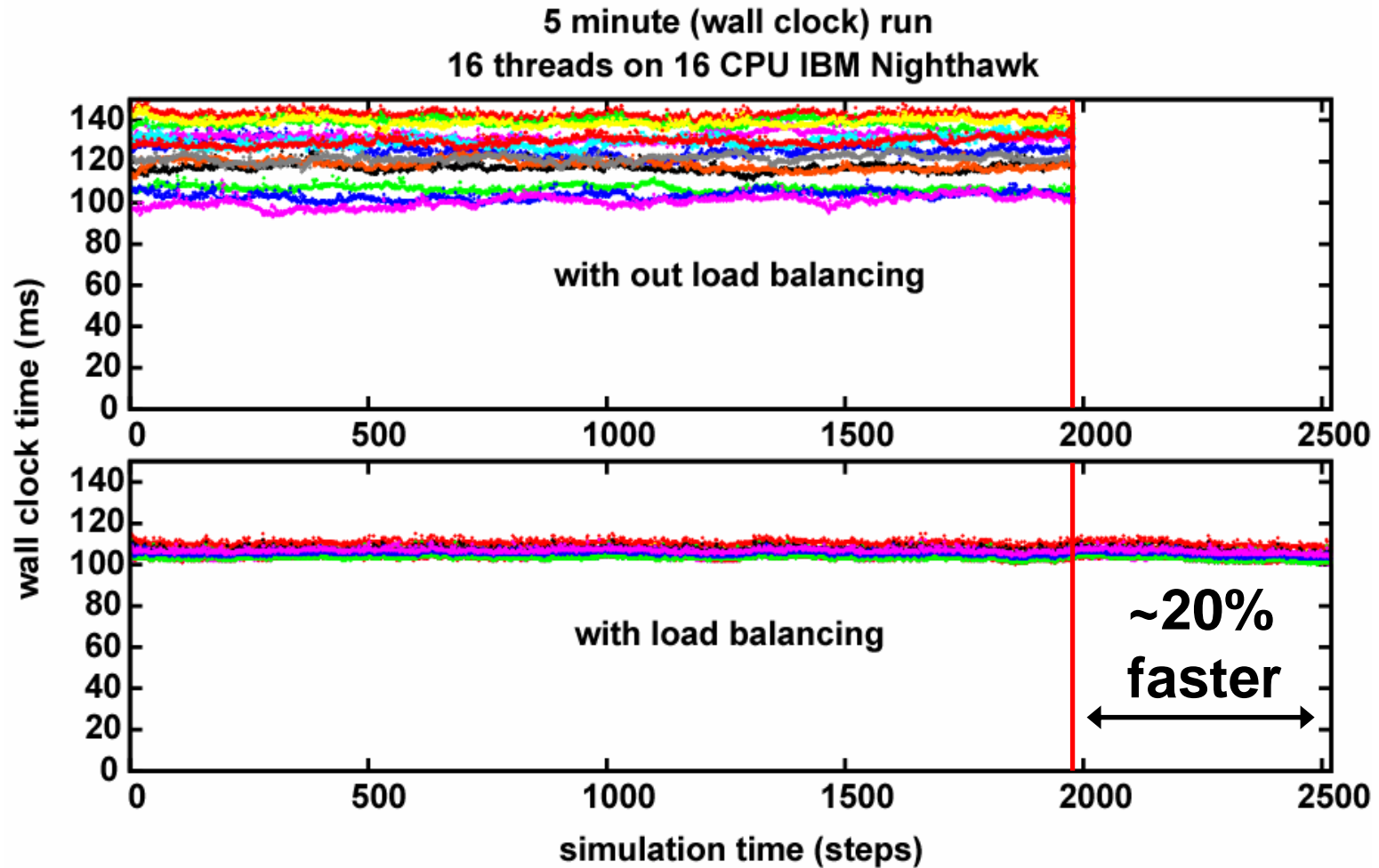
- **NERSC DOE INCITE award**
  - 2,000,000 + hours
  - 906 simulations of 151 protein folds on Seaborg



- One to two simulations per node (8 – 16 CPUs / simulation)
- Opportunity to tune *i/mm* for maximum performance

# Dynameomics

- Load balancing
  - Even distribution of non-bonded pairs to processors

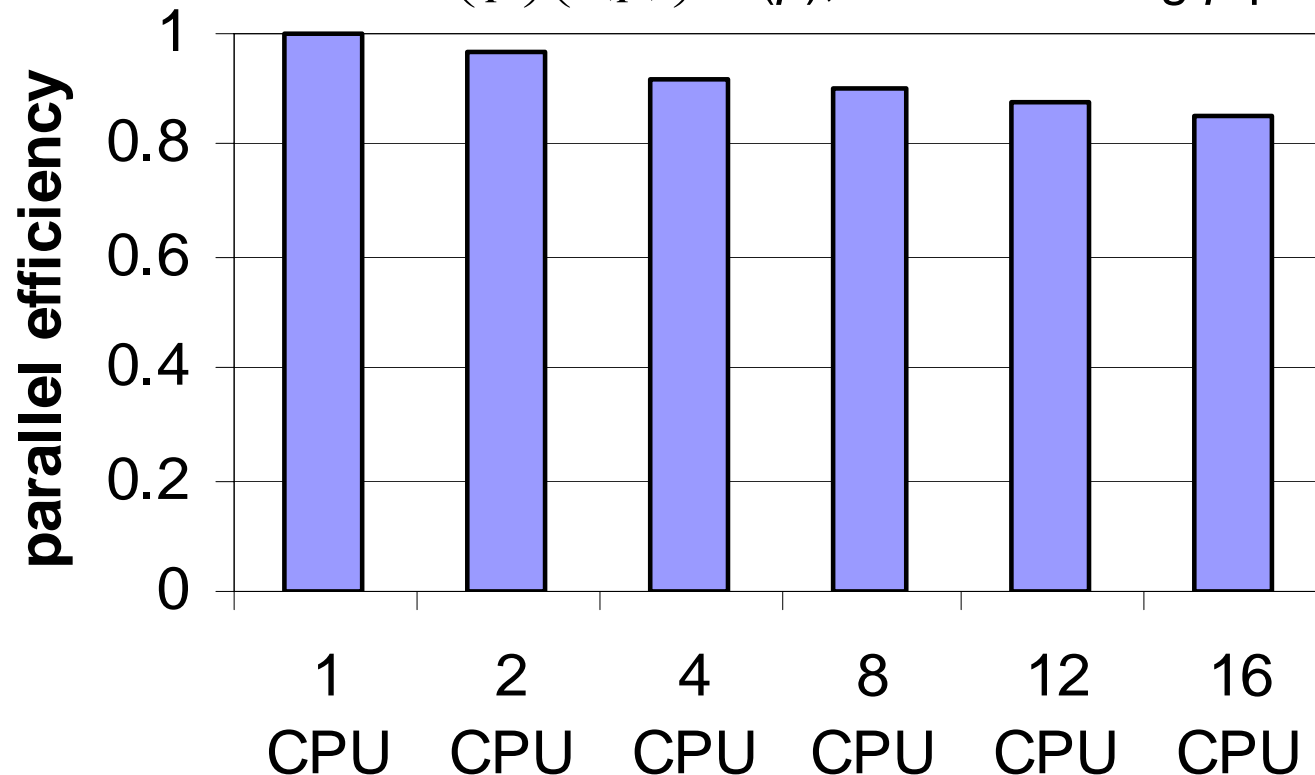


# Dynameomics

---

- **Parallel efficiency**
  - **Threaded computations on 16 CPU IBM Nighthawk**

parallel efficiency,  $e(p) = \left(\frac{1}{p}\right) \left(\frac{t(1)}{t(p)}\right)$   $p$ , number of processors  
 $t(p)$ , run-time using  $p$  processors





# Dynameomics

---

- **Simulate representative from top 151 folds**
  - 151 folds represent about 75% of known proteins
    - ~ 11  $\mu$ s of combined sim. time from 906 sims!
    - ~ 2 terabytes of data (w/ 40 to 60% compression!)
    - ~ 75 / 151 have been analyzed
    - Validated against experiment where possible

# Dynameomics

---

- **Now what?**
  - **Simulate the top 1130 folds (>90%)**
    - **More CPU time**
  - **Share simulation data from top 151 folds w/ world:**

**[www.dynameomics.org](http://www.dynameomics.org)**

- **Coordinates, analyses, available via WWW**
  - **MicrosoftSQL database w/ On-Line Analytical Processing (OLAP)**
    - **End-user queries of coordinate data, analyses, etc.**
- **Data mining**
  - **More CPU time, clever statistical algorithms, etc.**

# Acknowledgements

---



- DOE / NERSC's INCITE (David Skinner, *et al*)
- NIH
- Microsoft, Inc.
- Structures rendered using Chimera, Molscript, Raster3D & PyMOL